

Sustainable NH₃ Production: New Tech and Technoeconomic Considerations

Joshua M. McEnaney

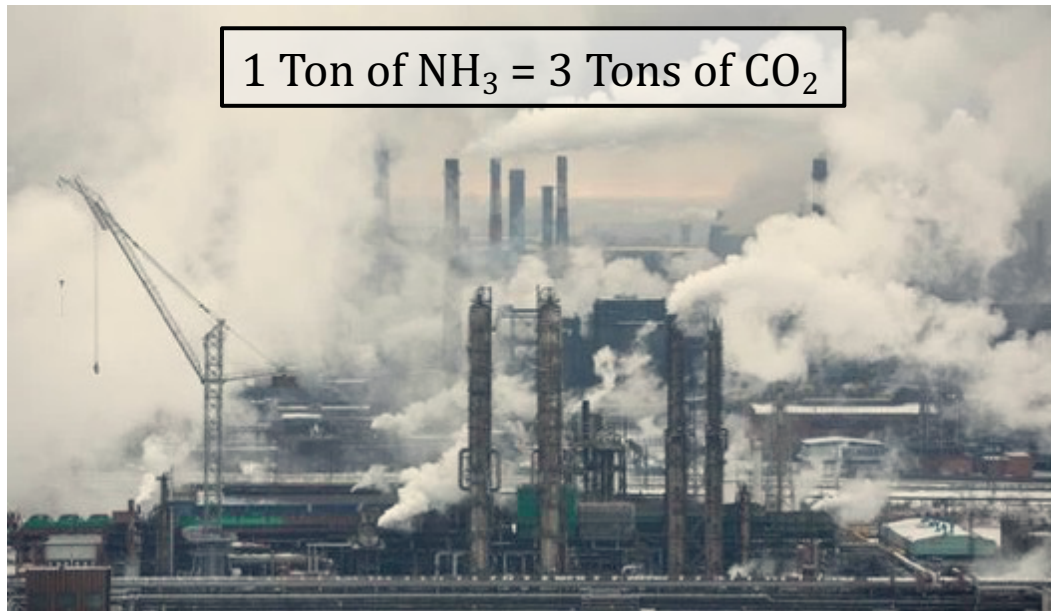
Postdoctoral Researcher: Stanford University
Co-Founder: Nitricity

Ammonia Energy Society Session
AICHE National Meeting, Fall 2019, Orlando, FL

Overview

Background

Nitrogen fertilizer has been produced the same centralized way for 100 years via a carbon- and capital-intensive process.



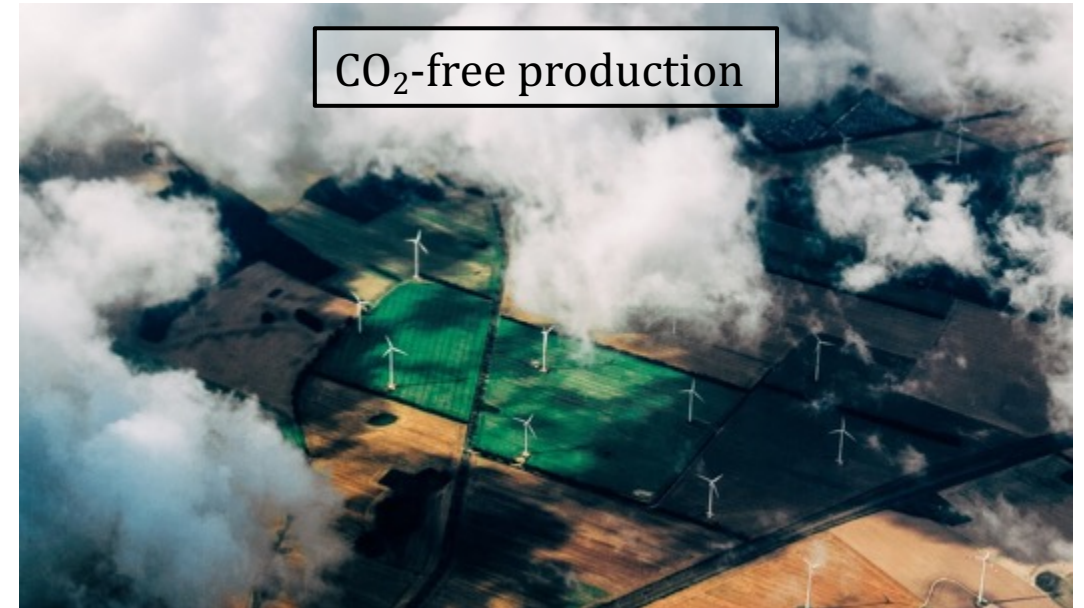
1 Ton of NH_3 = 3 Tons of CO_2

Billion-dollar, city-sized facilities

Image: Forbes.com

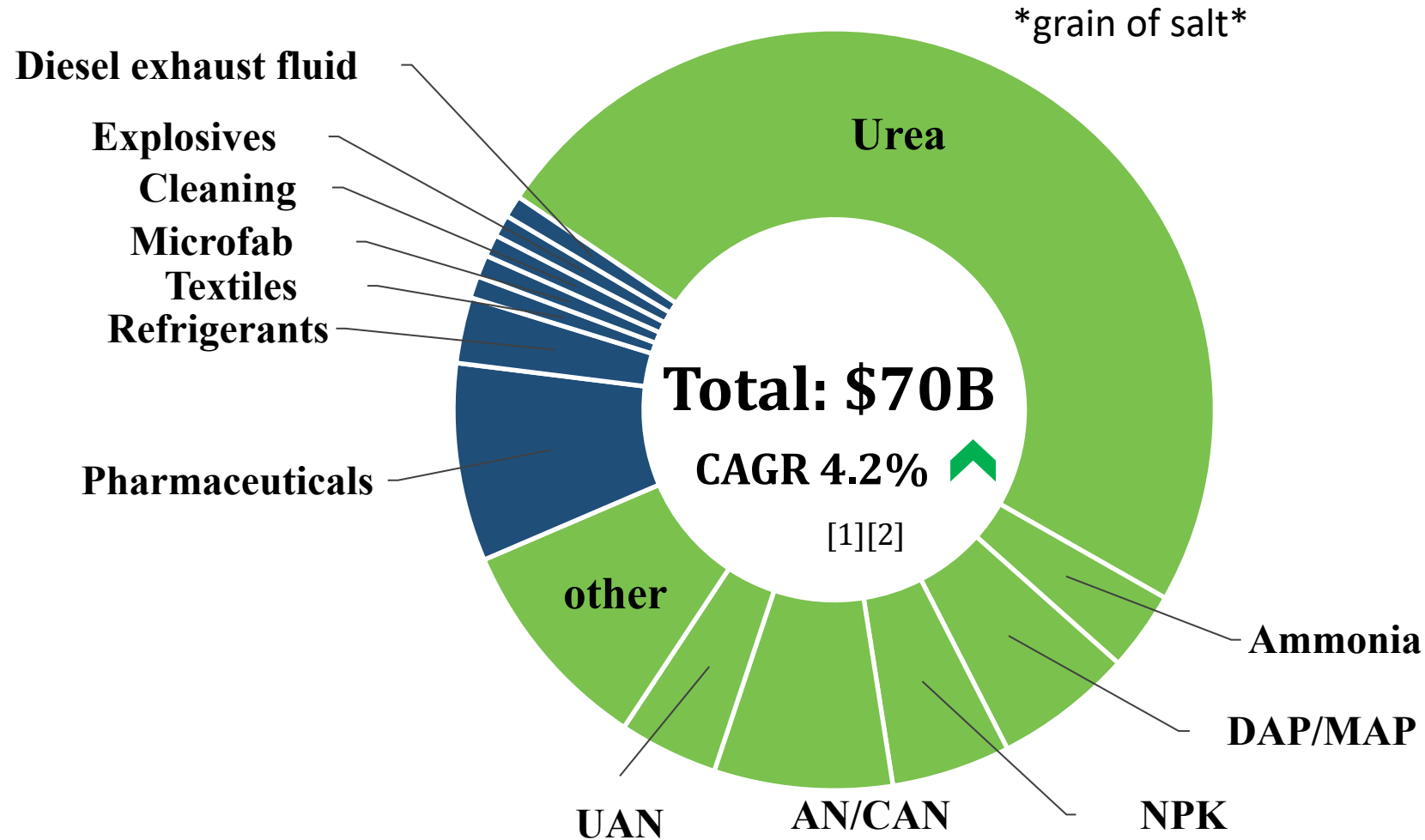
Goal

Carbon-free nitrogen closer to the end user using only air, water, and renewable electricity.



CO_2 -free production

Ammonia market | overview



165M tons of ammonia
produced in 2017 [3]

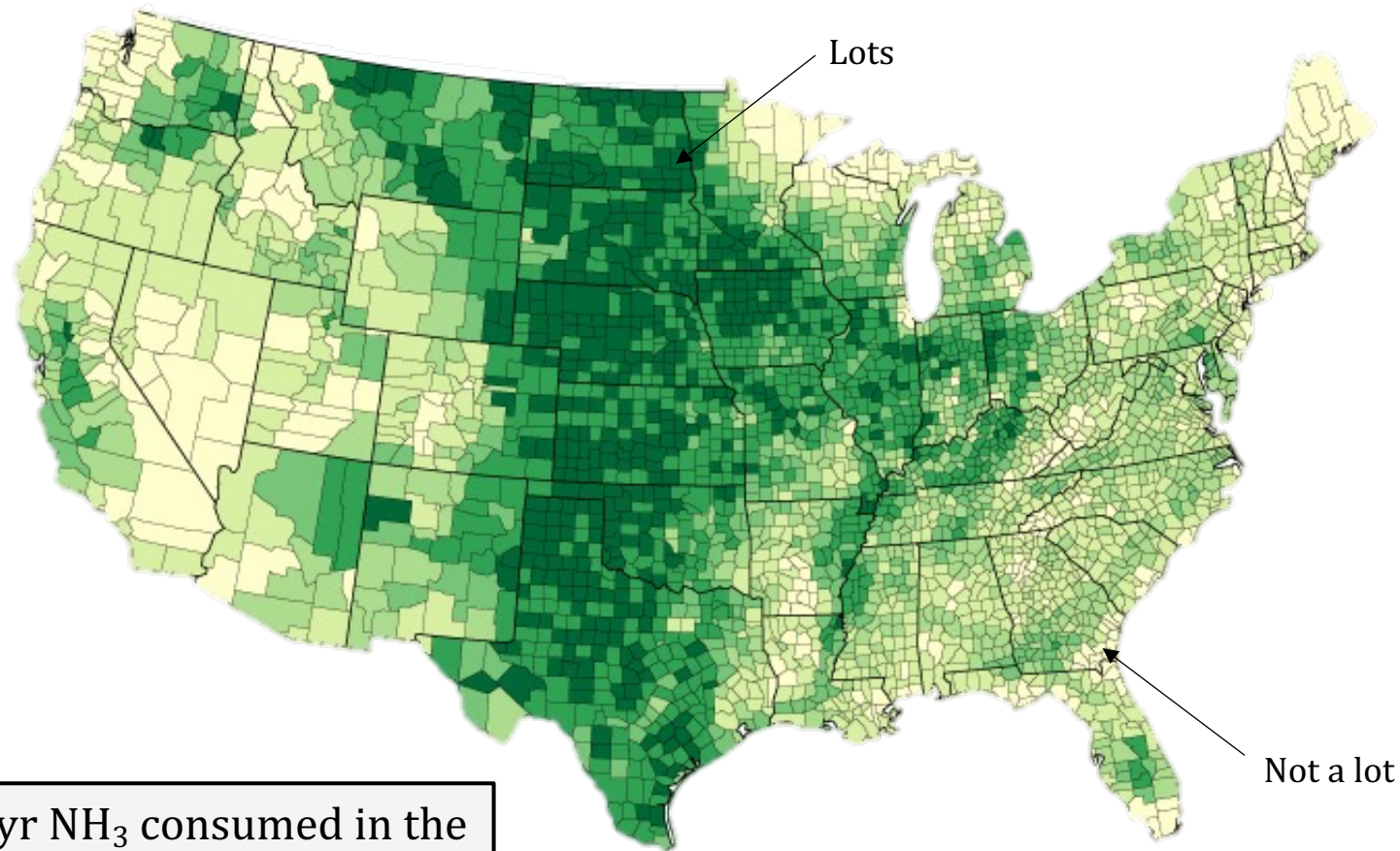
Fertilizer >75%
Industrial

[1] Yara Annual report 2017

[2] Marketwatch.com

[3] USGS annual ammonia report 2017

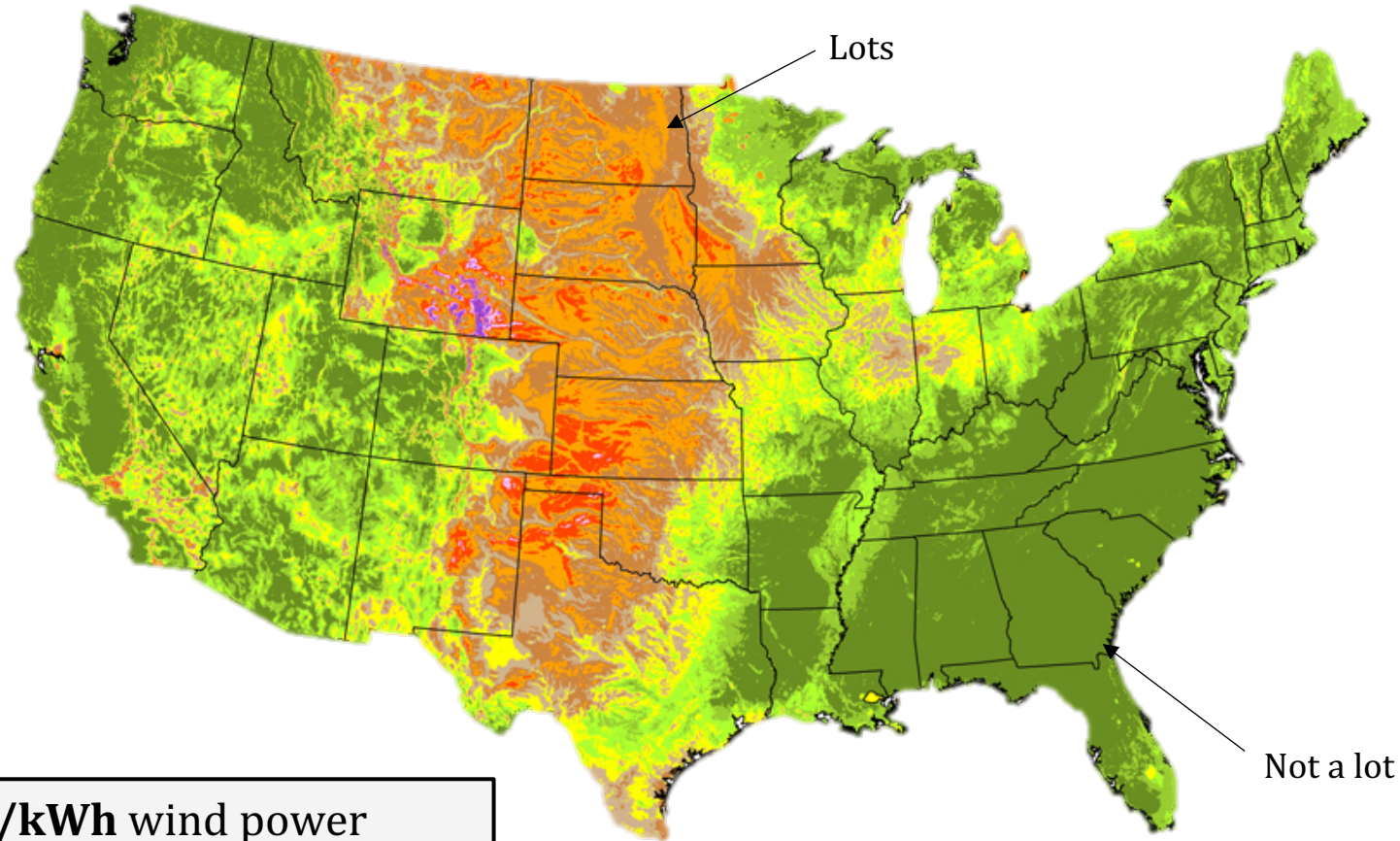
Farmland in the U.S.



7,000,000 tons / yr NH_3 consumed in the
Midwest alone

Source: USDA

Wind Resources in the U.S.

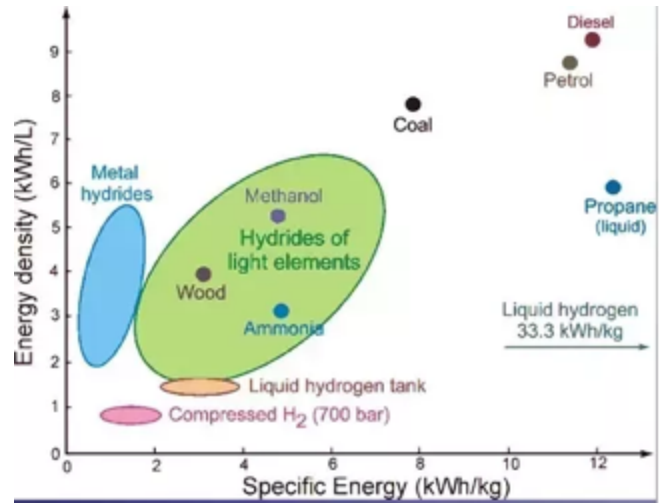


\$0.02-\$0.03/kWh wind power
in the Midwest

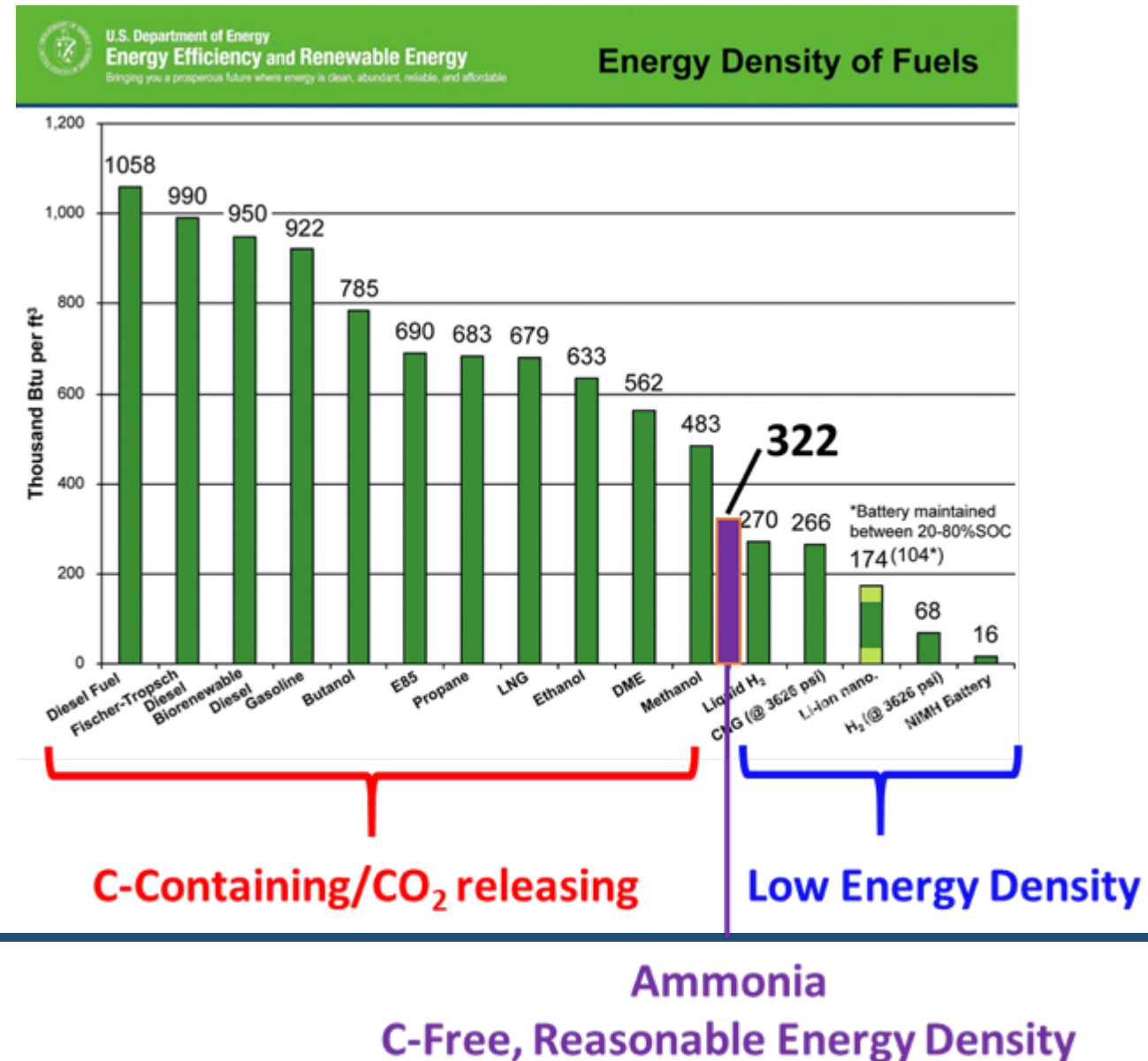
Source: NREL

Ammonia to Feed and Fuel

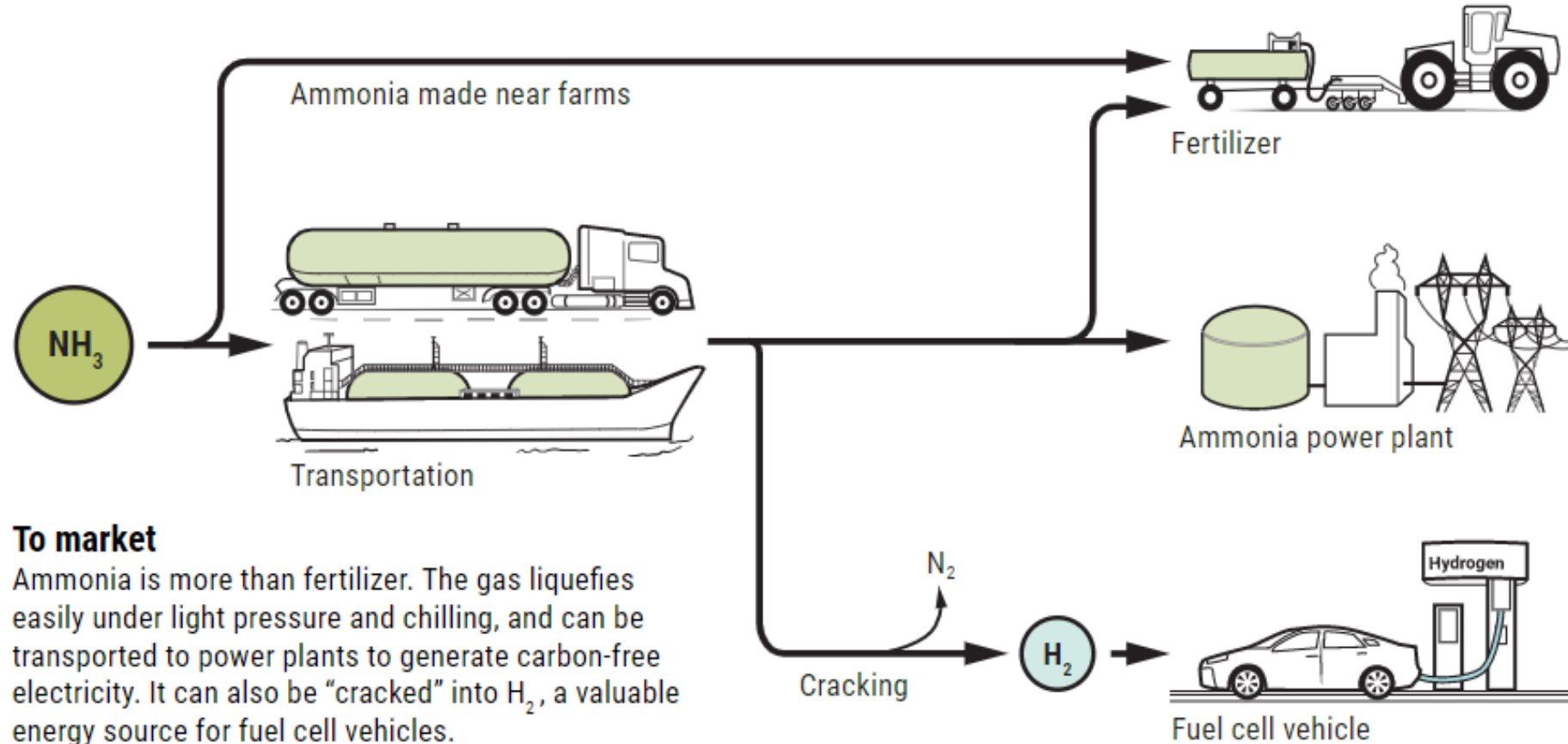
Growing idea to use NH₃ as a fuel



- Clean, carbon free emissions
$$2\text{NH}_4 + 2\text{O}_2 \rightarrow \text{N}_2 \text{ and } 4\text{H}_2\text{O}$$
- Proven safety record with strong infrastructure
- Flexible end-use



NH₃ Energy Economy



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Technology

**Global technology
survey**

**Interviews and
experimental testing**

**Proprietary TEA
portfolio**

NH₃ Technology Landscape

- Interviews and study of many groups around the world
- Included a study of key applied parameters and key risks
- Compiled and categorized technology into 3 major new-tech groups



[1] NH₃ Report Jesse S.

NH₃ Technology Landscape

- Techno-economic analysis indicates that e-chem presents an exciting technology pathway.



[1] NH₃ Report Jesse S.



**Improved
Haber-Bosch**

Moderate-pressure, CO₂ intensive

**50-150 Bar,
Still difficult to scale down**

**E-chem +
Haber-Bosch**

Requires intermediate H₂ production

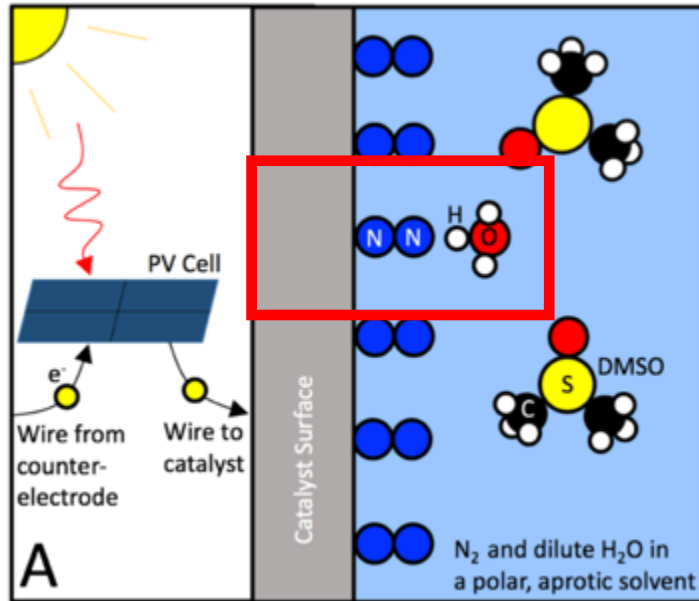
**CapEx hurdles including H₂ compression and storage.
H₂ production cost remains high. Promising**

**Electrochemical
Methods**

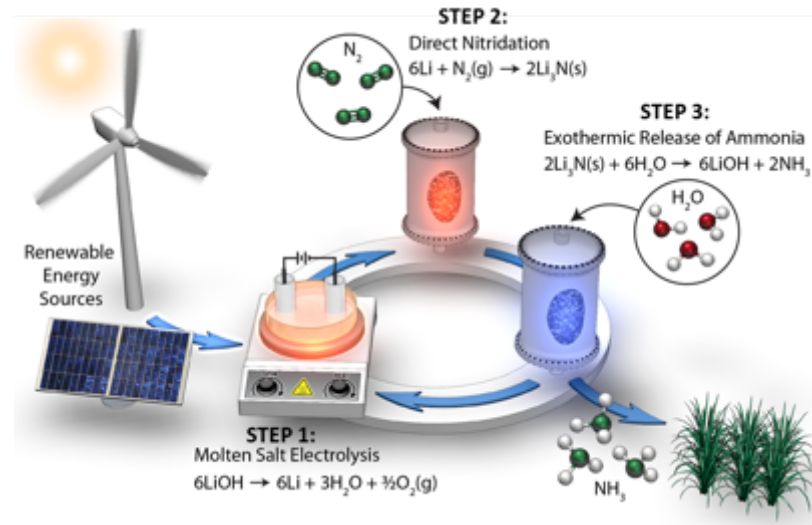
Exciting opportunity!

Less tech-ready

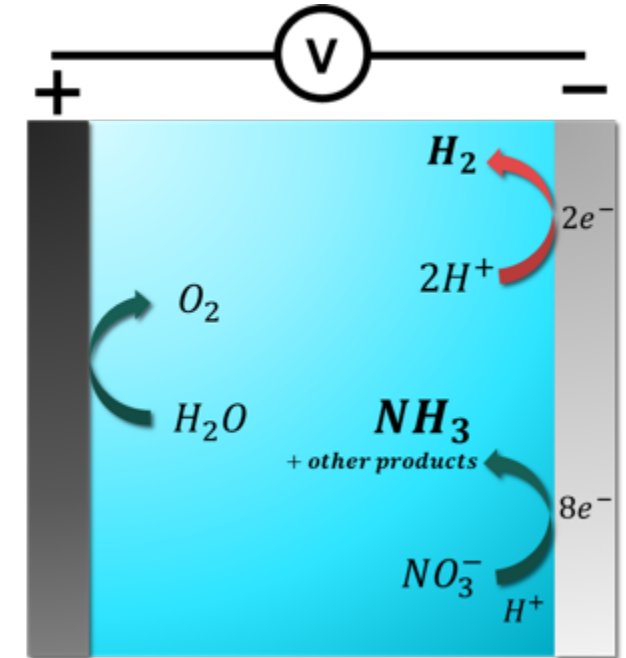
3 Electrochemical Tech Highlights



“Direct electrochemical”

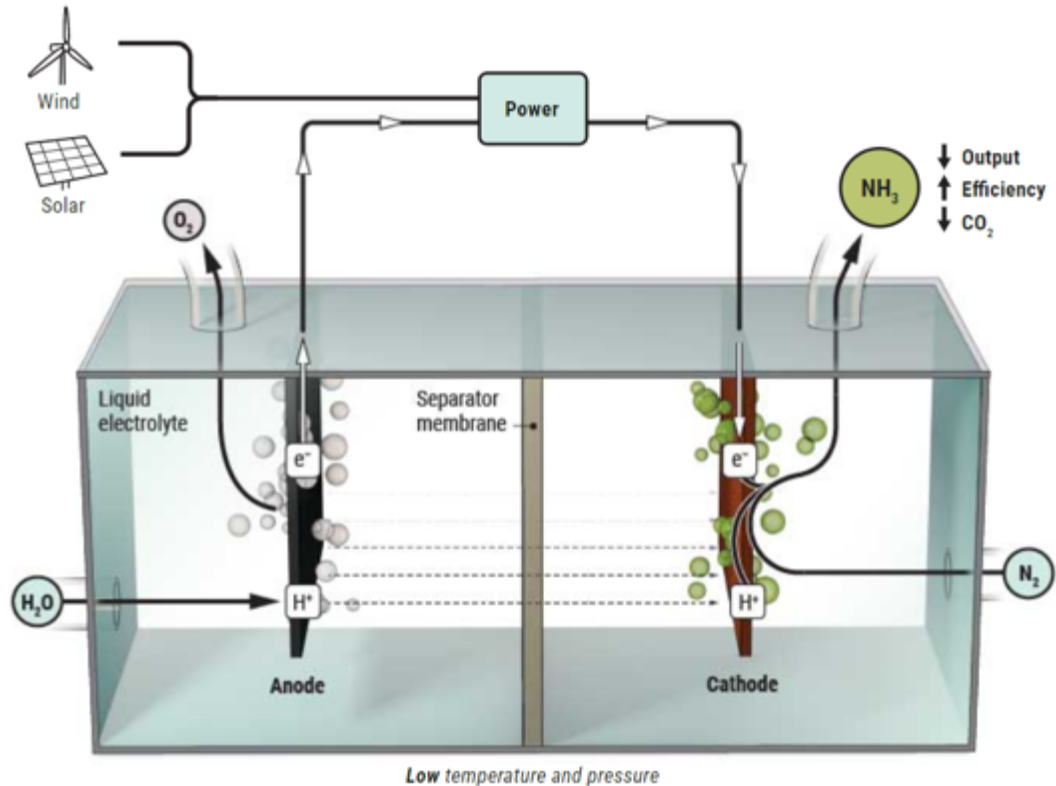


Electrochemical metal cycling



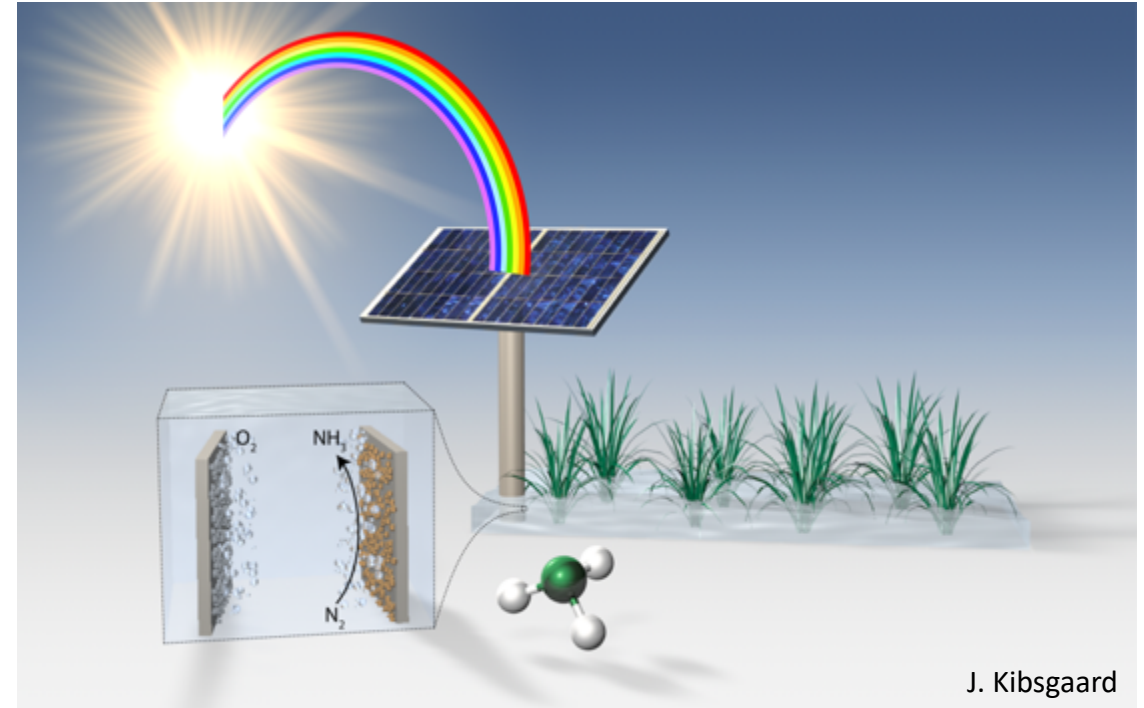
Nitrate reduction

New Tech Highlights: **Direct e-chem**



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The "Ideal"

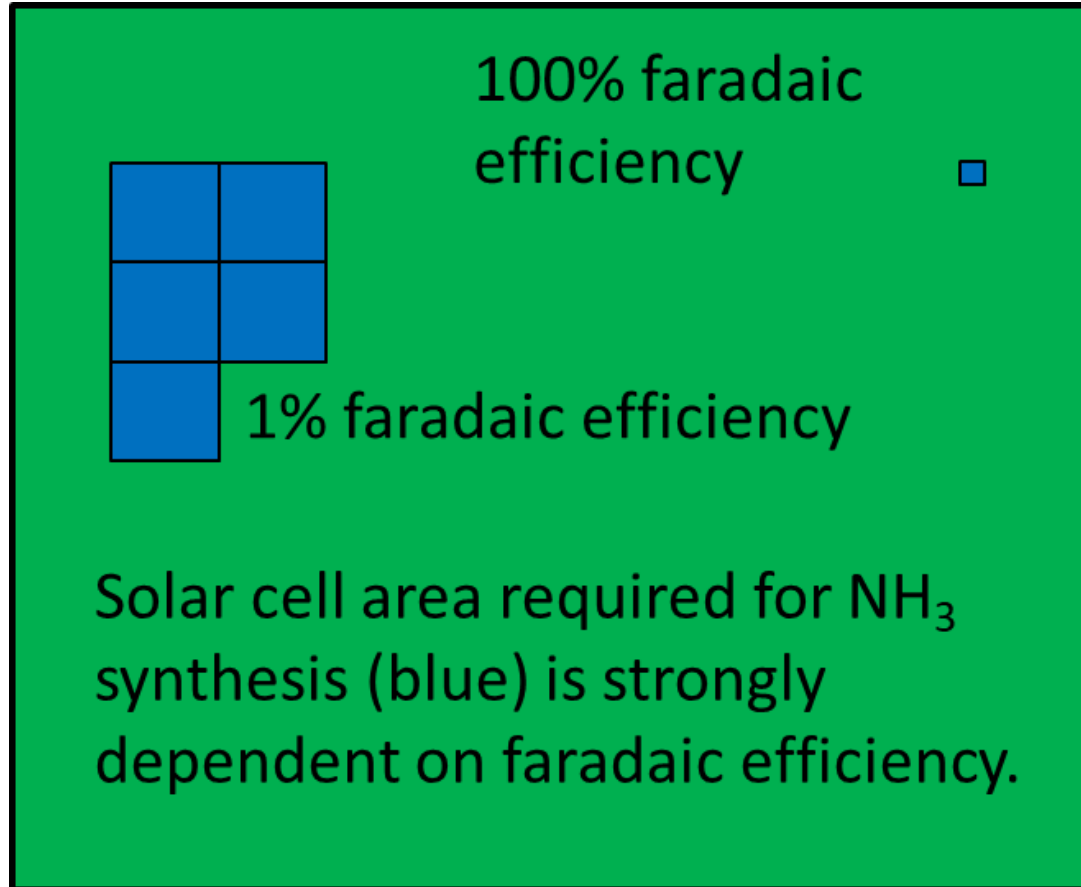


J. Kibsgaard

- Sustainable production from air, water, electricity
- Produced when sun is shining, as needed
- Ambient pressure and temp conditions possible
- Low capital cost gives entitlement for broad use

New Tech Highlights: **Direct e-chem**

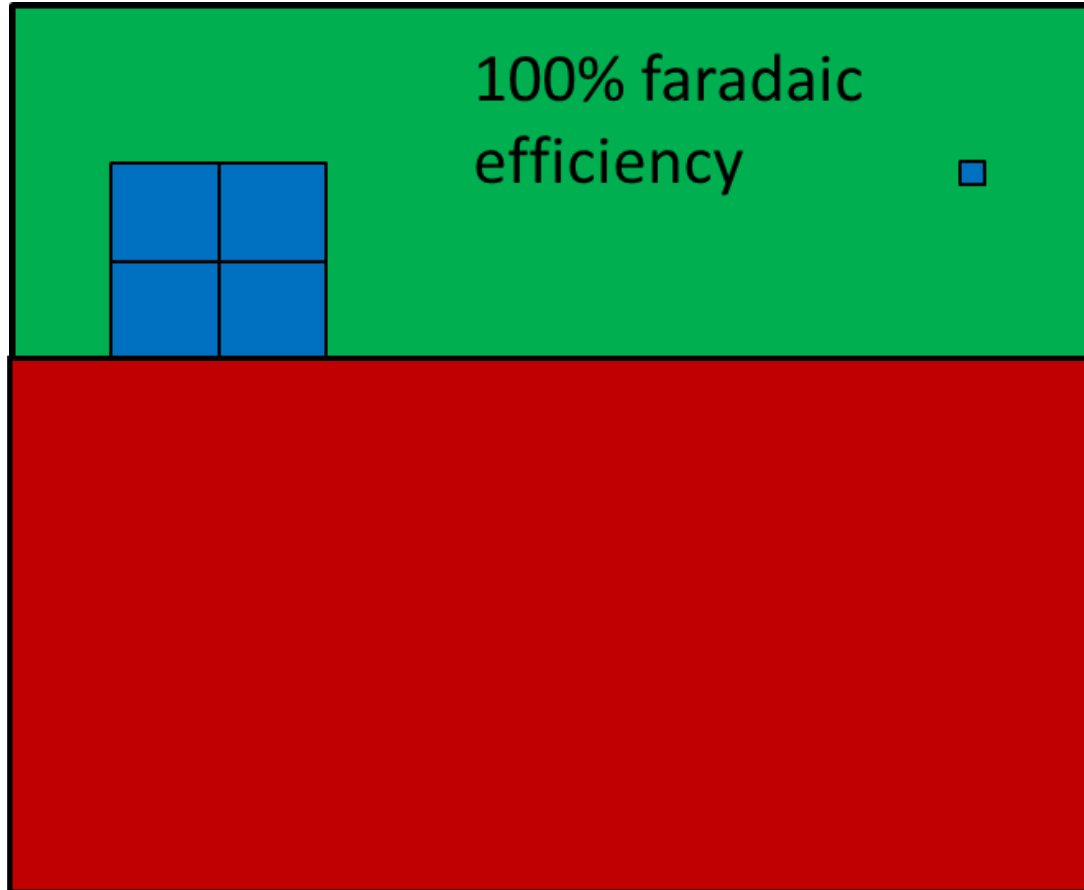
Average farm



- If efficient, requires very little solar area per farm land
- However, most efficient reports operate at only a few $\mu\text{A}/\text{cm}^2$ current densities... (or several $\text{mg NH}_3/\text{m}^2 \text{ h}$)

New Tech Highlights: **Direct e-chem**

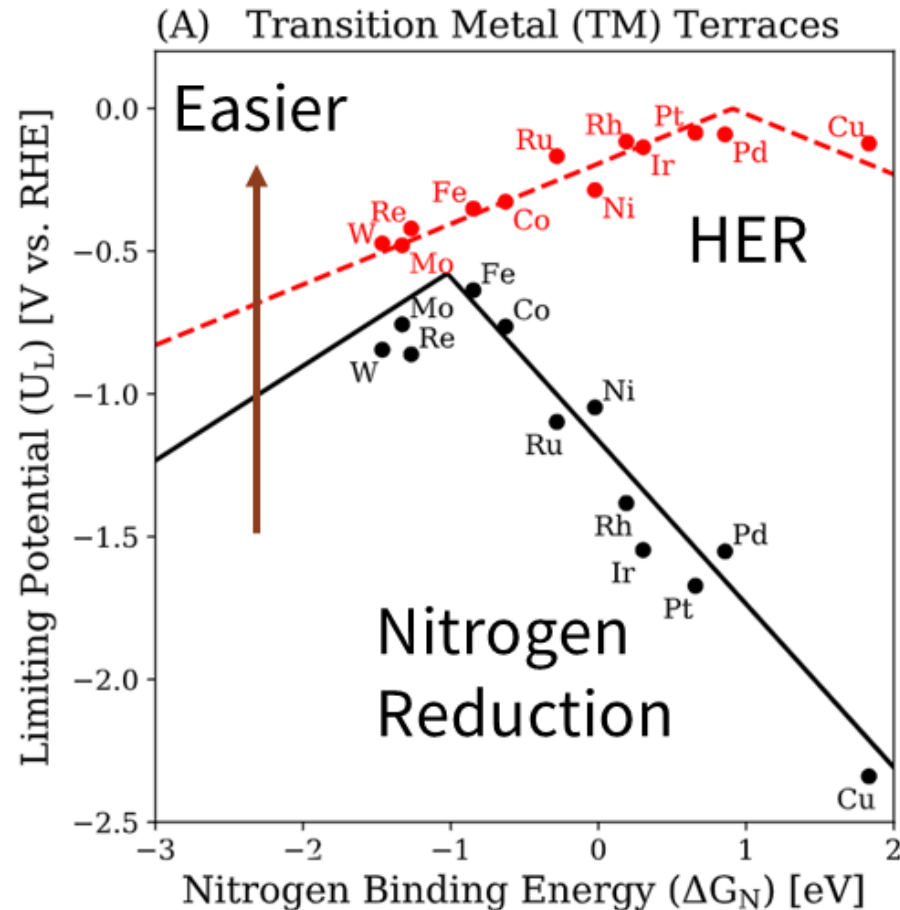
Average farm



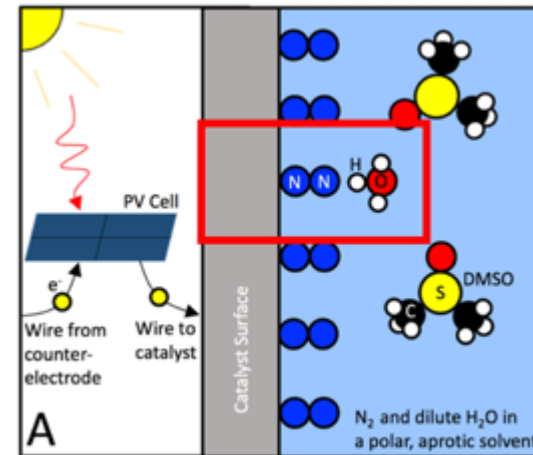
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— Approx. geometric area of catalyst required at this rate
Equates to high capital costs – rate must improve!

New Tech Highlights: Direct e-chem

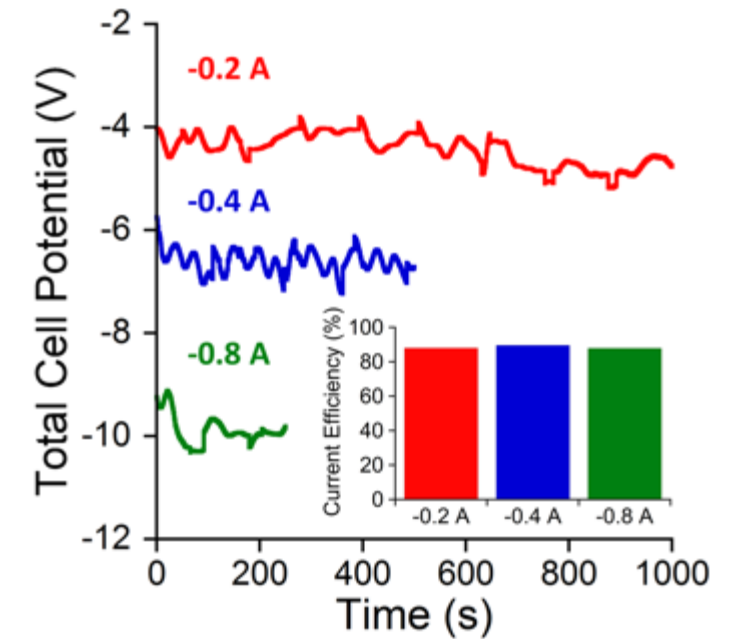
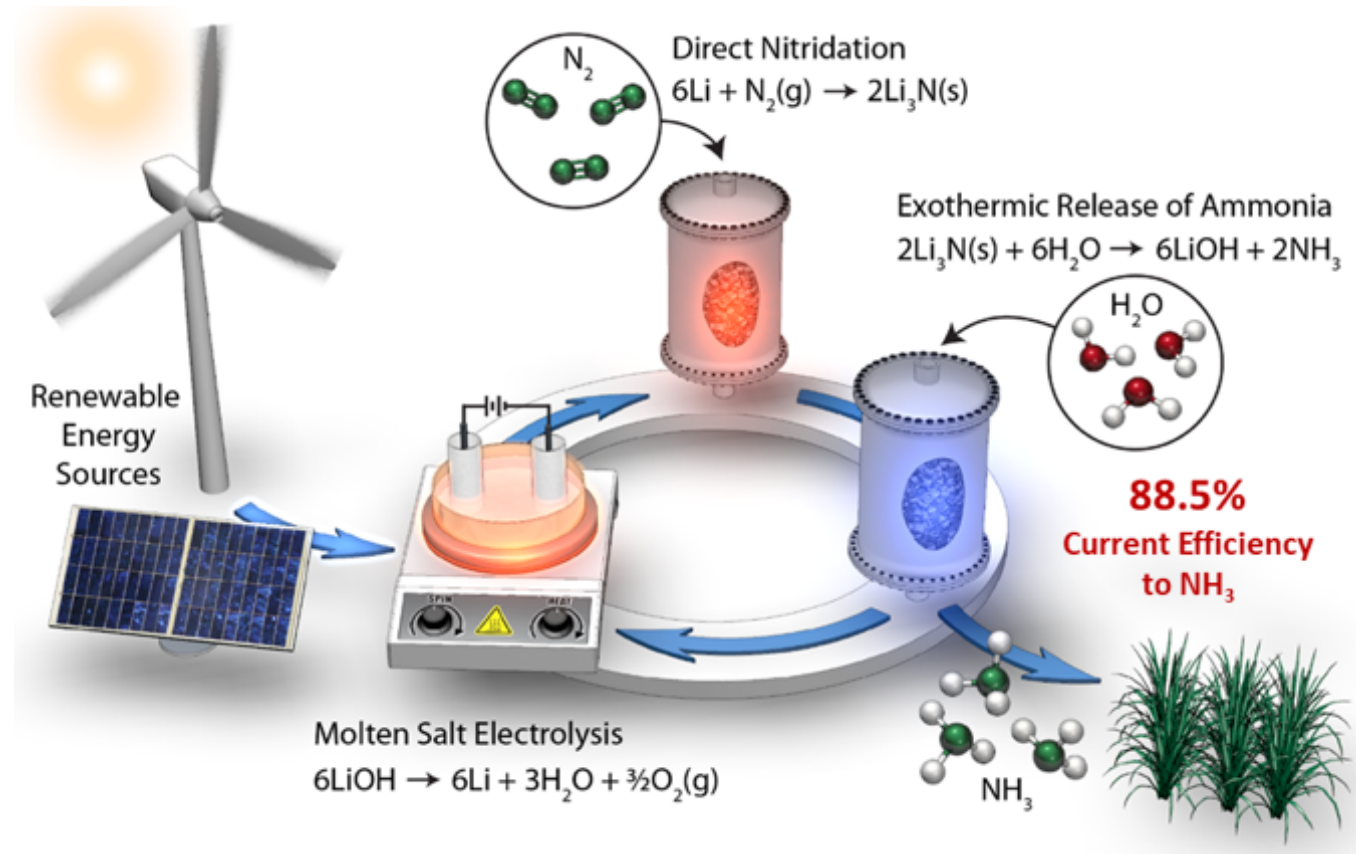


- Hydrogen evolution dominates with water present
- Literature prone to false positives
- Most less than 1% FE and/or very low current densities ($<1\text{mA}/\text{cm}^2$)

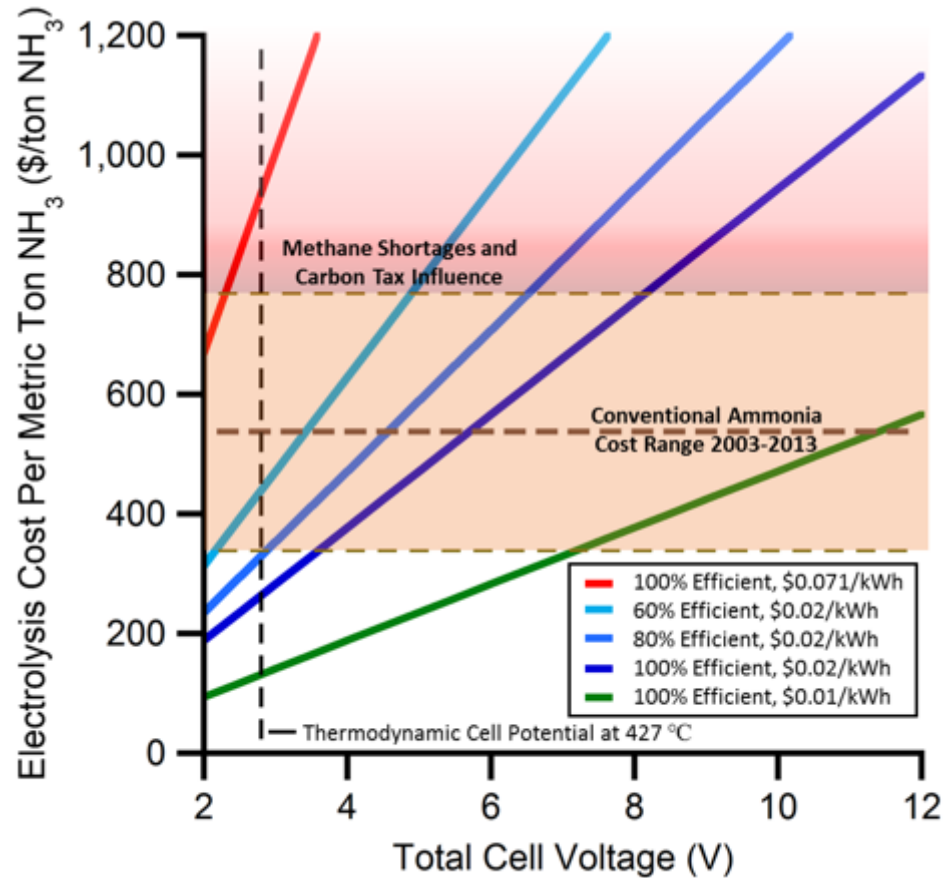


- Limiting protons can improve efficiency
- Stability not yet a focus
- Low “tech-readiness”, long term promise

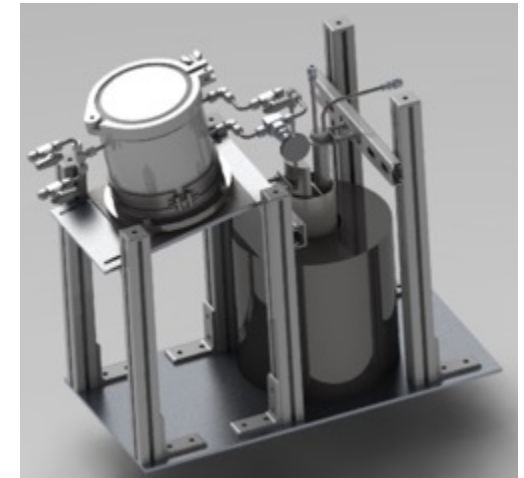
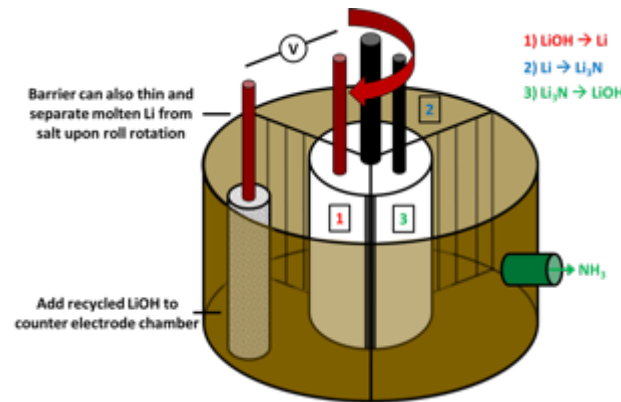
New Tech Highlights: Electrochemical metal cycling



New Tech Highlights: Electrochemical metal cycling



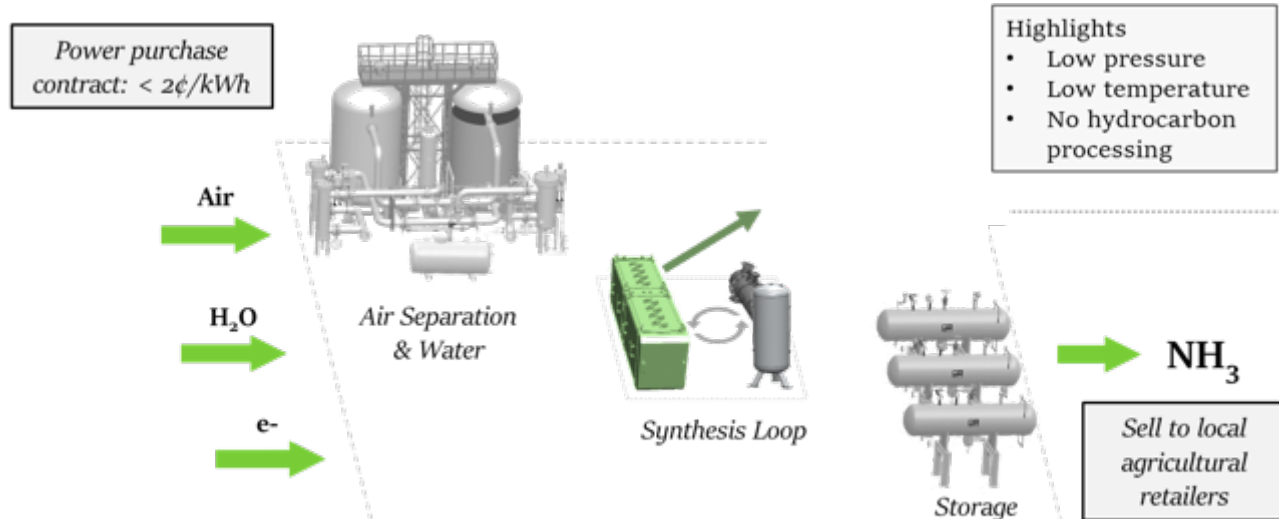
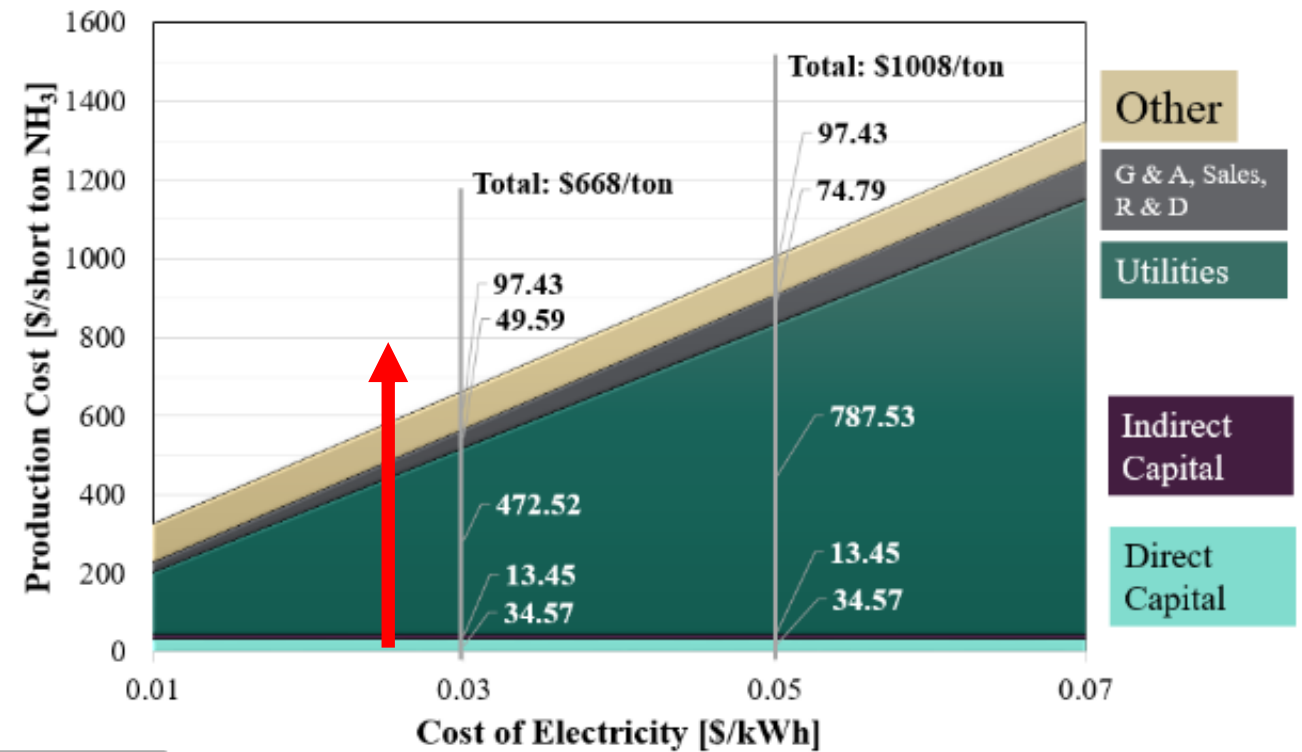
- 1000x the rate of “direct electrochemical” methods today (2 A)
- ~90% current efficiency to NH_3 product demonstrated
- Ambient pressure, moderate temperature (350 C)
- Designed and built prototypes for continuous operation and practical implementation



- Electricity cost of electrolysis, not prohibitive with cheap electricity

Electrochemical metal cycling

- Detailed plant modeling and TEA



- 10 TPD scale
 - ~~10 year depreciation~~ → 2 year depreciation
 - Not too far off – need dirt cheap electricity
- Li cell stability requires innovation to decrease CapEx**

[1] McEnaney J. et al. (2017). Energy Environ. Sci 2017. 10, 1610.

[2] Stratasys

New Tech Highlights: Nitrate reduction

Nitrate waste is an environmental pollutant

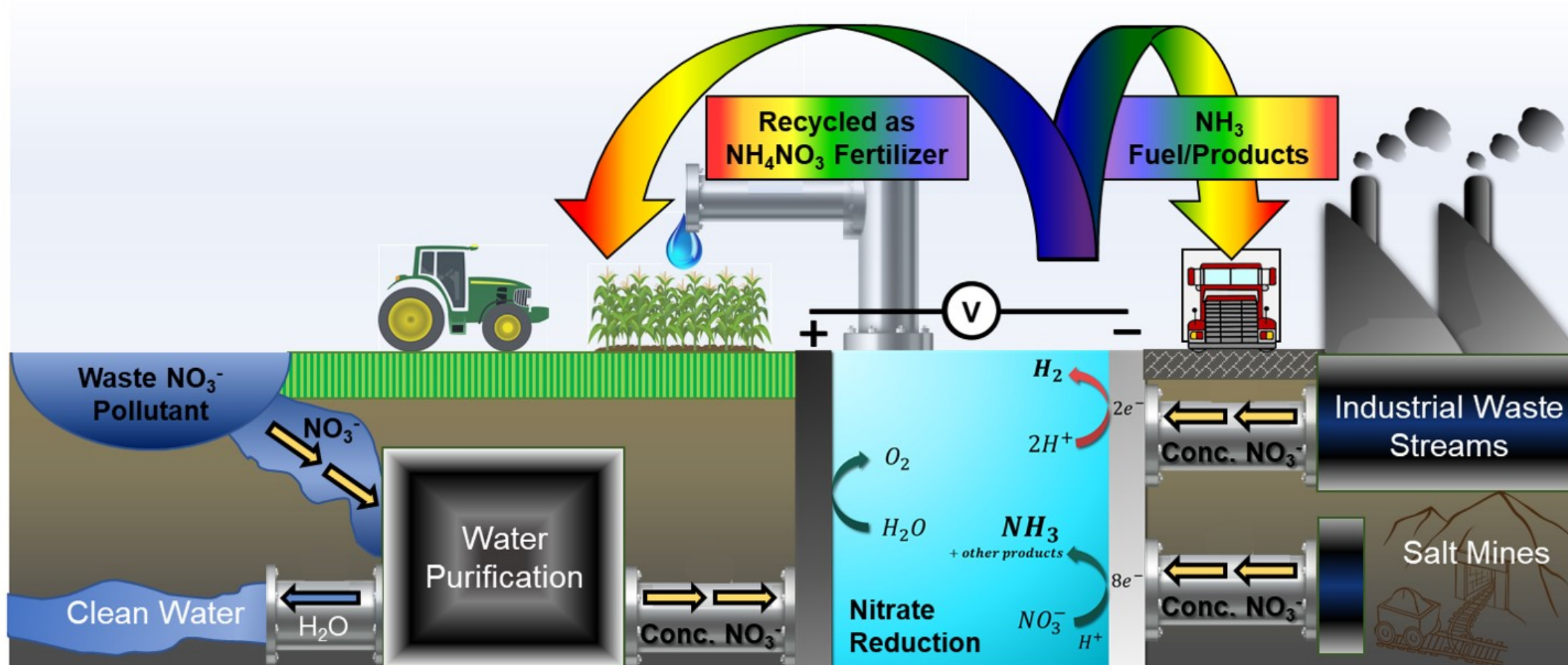
- Algal blooms/Habitat loss
- Birth defects/Health issues
- Significant denitrification research to form N_2 gas...

Nitric acid is made from NH_3

- Source of NH_4NO_3 fertilizer
- N_2O byproduct is a potent GHG (300x CO_2)
- 118 MmTCO₂e GHG emissions
- ~1% of all non CO_2 e from HNO_3

What if we could turn waste nitrates into a useful product?

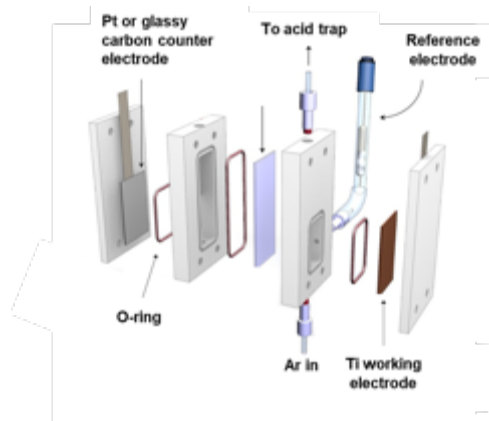
New Tech Highlights: Nitrate reduction



New Tech Highlights: Nitrate reduction

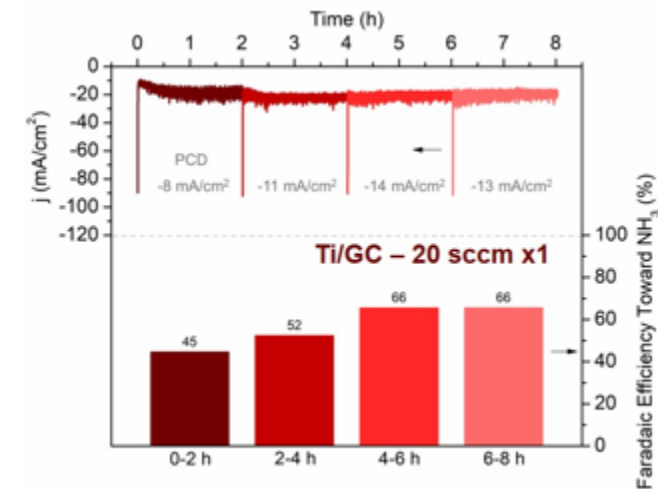
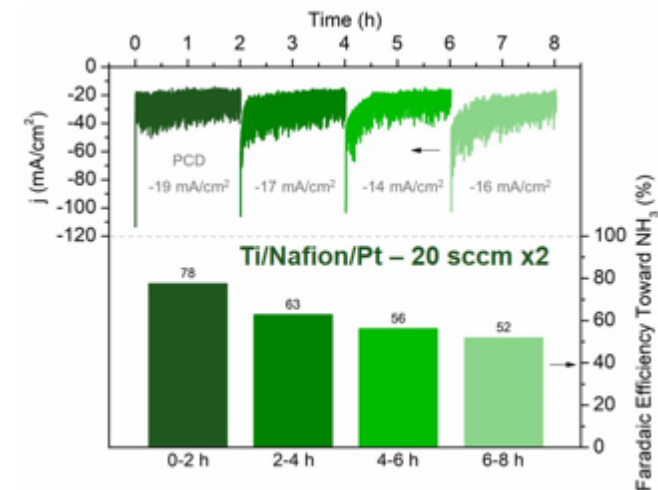
Electrolyte engineering for selectivity to NH_3 : Success!

Literature Comparison				
Cathode	Conditions	Max Faradaic Efficiency to NH_3 Reported	Partial Current Density to NH_3 (mA/cm^2)	Reference
Fe Cu Ti	100 mg/L (1.2 mM) NaNO_3 -20 mA/cm^2 applied Neutral pH	11 % 3.2 % 2.5 %	-2.2 -0.6 -0.5	<i>Electrochimica Acta</i> 2009, 54 (20), 4600-4606.
Al Bi Zn Sn In Pb	0.050 M NaNO_3 -1.8 V vs Ag/AgCl Variable pH (0.16-13.7)	12 % <10 % <10 % <10 % <10 % <10 %	N/A	<i>Environmental Technology</i> 2013, 34 (3), 373-381.
Ti	50 mg/L (0.6 mM) NaNO_3 -38 mA/cm^2 applied Neutral pH	5.6 %	-2.1	<i>J Electroanal Chem</i> 2016, 782, 270-277.
Sn	0.05 M NaNO_3 (2 M NaCl) -2 V vs Ag/AgCl Neutral pH	14.5 %	~-5	<i>Electrochimica Acta</i> 2007, 52 (23), 6412-6420.
Rh/graphite	1 M NaNO_3 (1 M NaCl) -4 mA/cm^2 Neutral pH	24 %	-1	<i>Electrochimica Acta</i> 2007, 52 (21), 6237-6247.
Ti	0.3 M KNO_3 0.1 M HNO_3 -1 V vs RHE Acidic pH	82 %	-22	This work

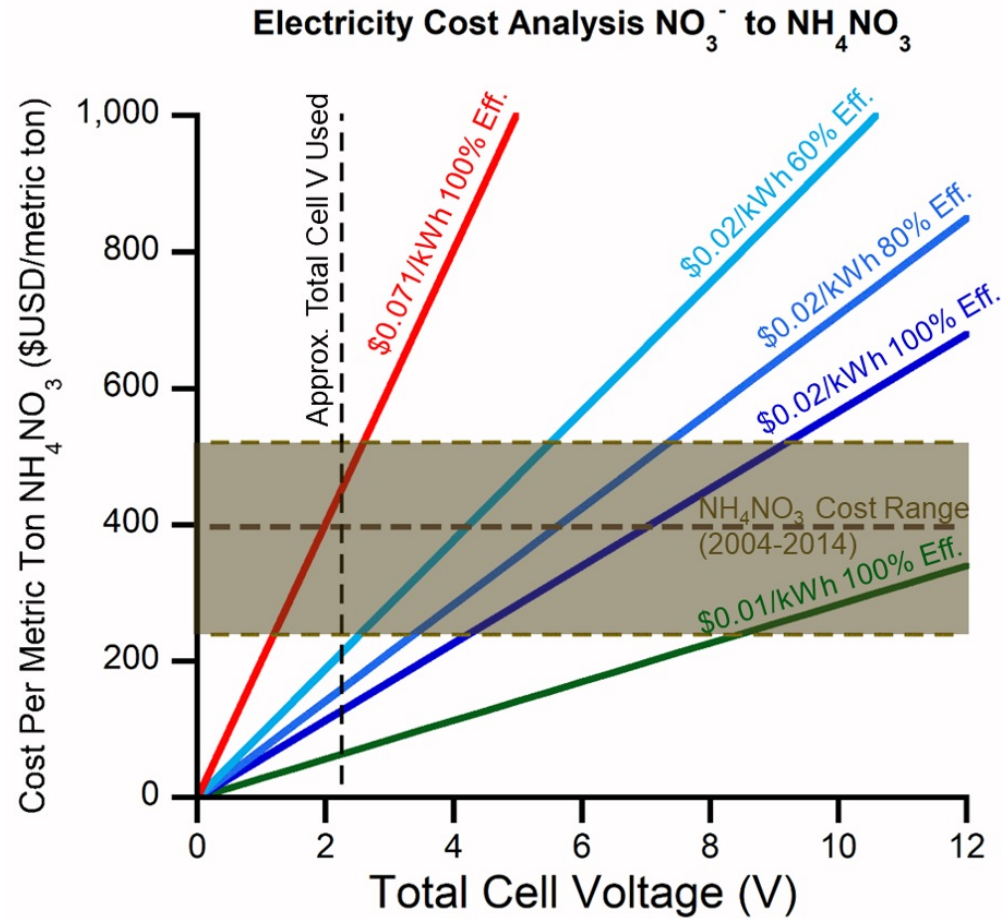


Literature denitrification focus (H_2 and N_2 products)

Optimization toward NH_3 production

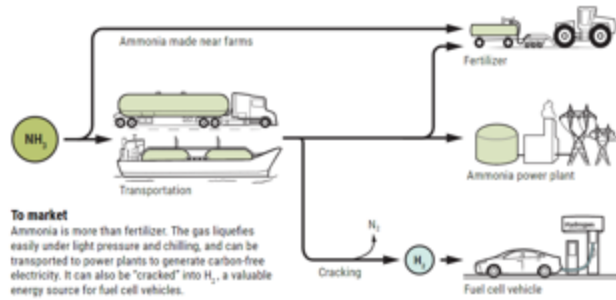


New Tech Highlights: Nitrate reduction

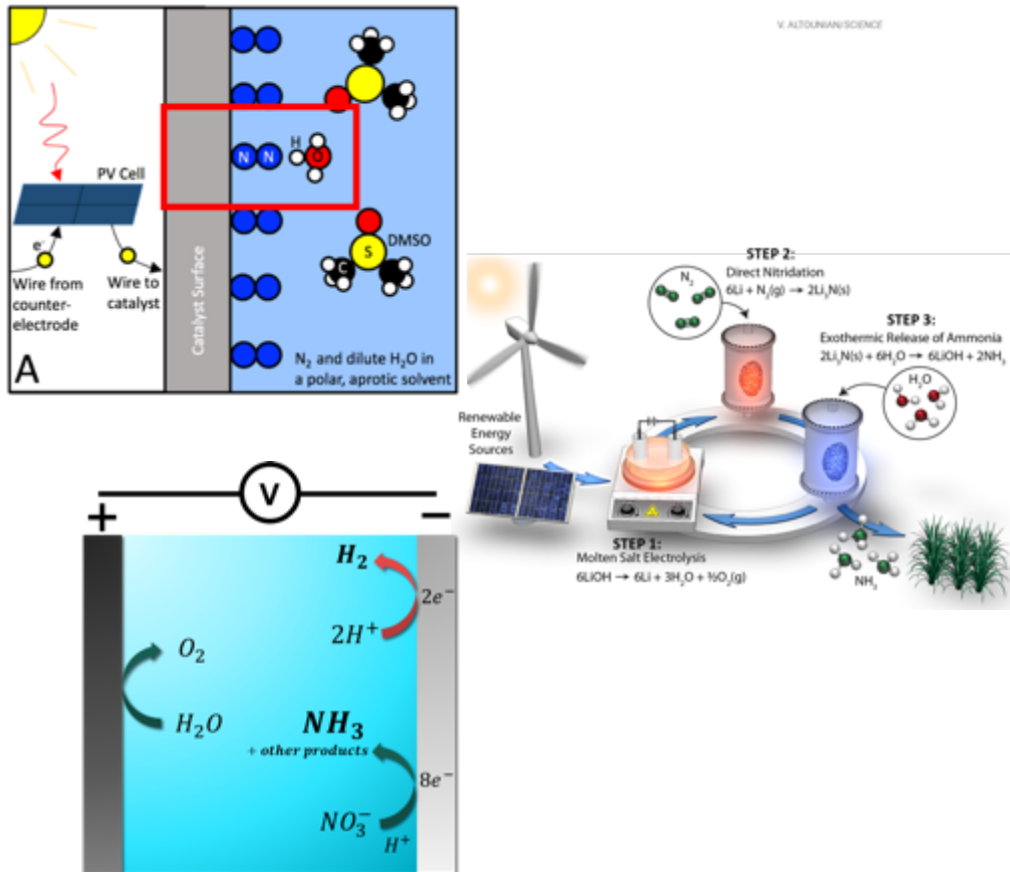


- Despite $8e^-$ reduction process, the electricity cost to produce NH_4NO_3 is not prohibitive
- Dependent on ability to concentrate waste nitrate source
- Need to mitigate electrode stability issues
- If electricity were cheap enough, could convert all to NH_3

Summary



New tech development to build NH_3 economy



- Direct Electrochemical (improvements)
 - Ideal, low-cost opportunity
 - Low rates and efficiencies
- Electrochemical Metal Cycling
 - Great rate and efficiency
 - Cell stability issues
- Electrochemical Nitrate Reduction
 - Great environmental opportunity
 - Electrode stability issues

Acknowledgements: V-Sustain NH₃ Team

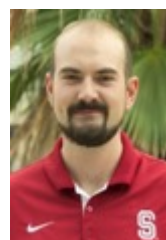
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- Jon Grant Baker
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- Stefano Mezzavilla
- Suzanne Z. Andersen
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- Viktor Čolić
- Jakob Kibsgaard
- Ifan E. L. Stephens
- Peter C. K. Vesborg
- Ib Chorkendorff

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SURFCAT
Surface Physics and Catalysis



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Brian Rohr | operations



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Breakthrough Energy Ventures

Engineering Scale-up



GJ Ia O'
Primus Power

Scientific Advisory Council



Jens Nørskov
Professor, DTU



Tom Jaramillo
Professor, Stanford University



A scenic landscape at sunset. In the background, a hill is topped with several wind turbines. The sky is filled with soft, orange and yellow clouds. In the middle ground, a small village with red-roofed houses is nestled among trees. The foreground is dominated by a vast field of yellow flowers, likely rapeseed, arranged in neat rows.

Thanh₃ks

www.enh3.com