

Basic co-firing characteristics of ammonia with pulverized coal in a single burner test furnace

Energy Engineering Research Laboratory
Central Research Institute of Electric Power Industry

Akira Yamamoto*, Masayoshi Kimoto, Yasushi Ozawa, Saburo Hara

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Objective of this study

Direct and blended combustion of ammonia in **pulverized-coal-fired power plant** is expected one of the promising technology regarding CO₂ reduction.

What is the significant challenge of ammonia as a fuel?

Ammonia (NH₃) contains N atom.

⇒ It can be **main source of fuel-NOx**.



If NOx emissions from combustion furnace increase;

- More quantity of NH₃ for flue-gas denitration (de-NOx)
- Improvement or addition of denitration equipment

In this study, we examined **the basic blended-combustion characteristics of pulverized coal and ammonia** using a single-burner combustion test furnace.

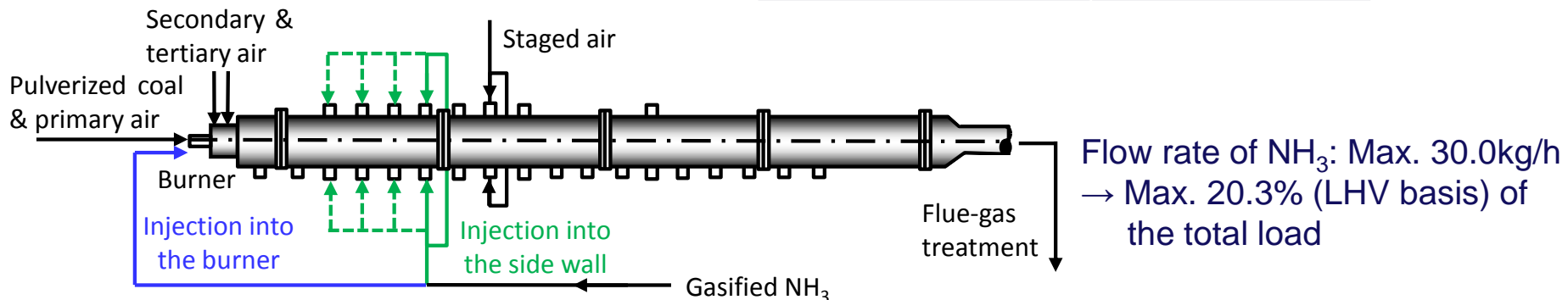
Outline of the test furnace

View of the single burner

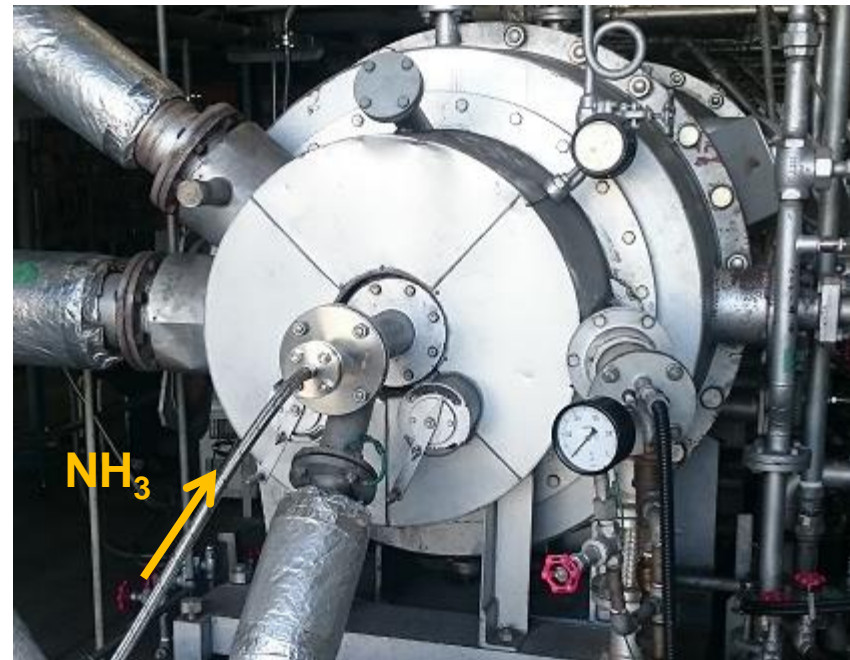
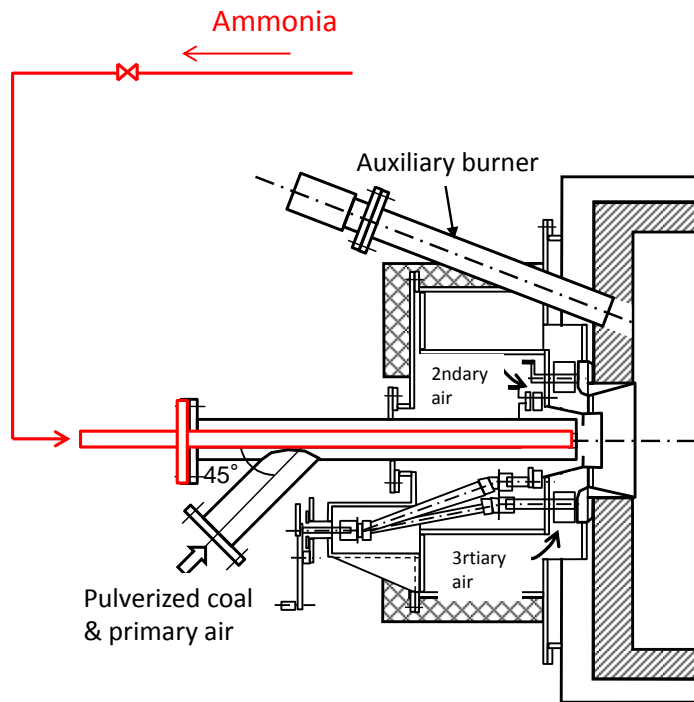


Typical conditions of the experiment

Item	Value
Combustion load (Coal consumption)	760 kW ($\dot{m} \approx 100$ kg/h)
Furnace size	$\Phi 0.85\text{m} \times 8\text{m}$
Set exhaust O_2	4.0 %
Staged air / Total air	30 %

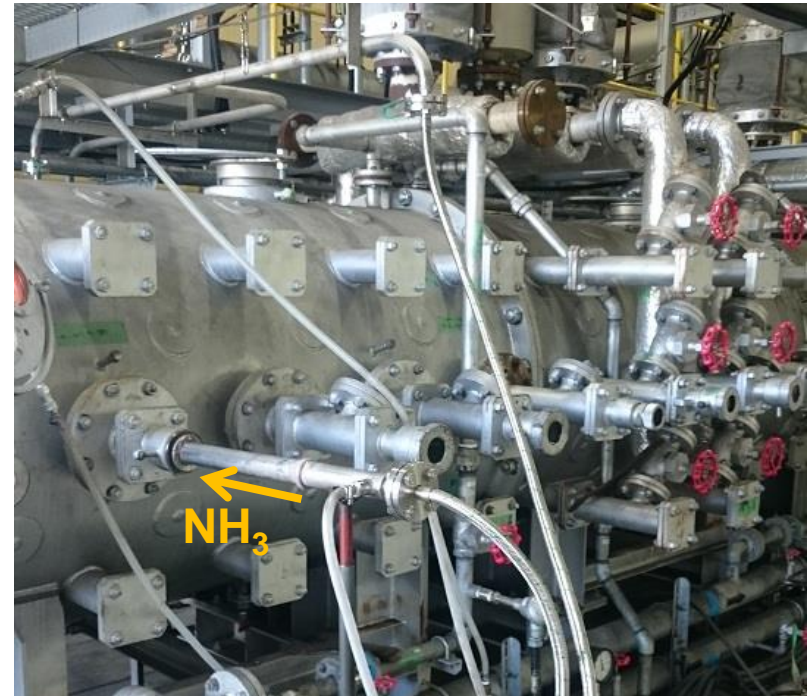
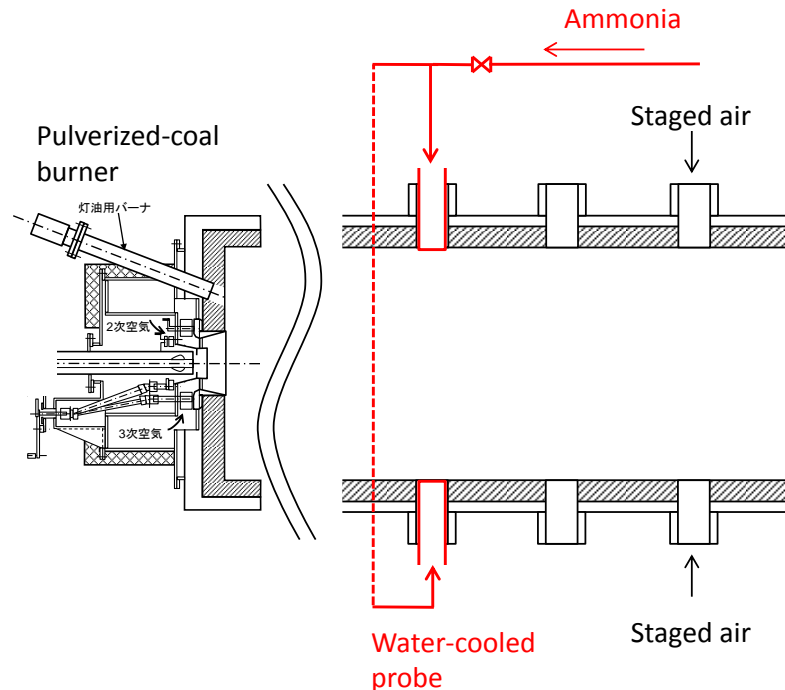


NH₃ injection into the coal burner



Ammonia injection pipe is inserted into the pulverized-coal burner.

NH₃ injection through the side port



Ammonia is injected into the pre-OFA zone through the measurement port.

Experimental conditions

Experimental parameters

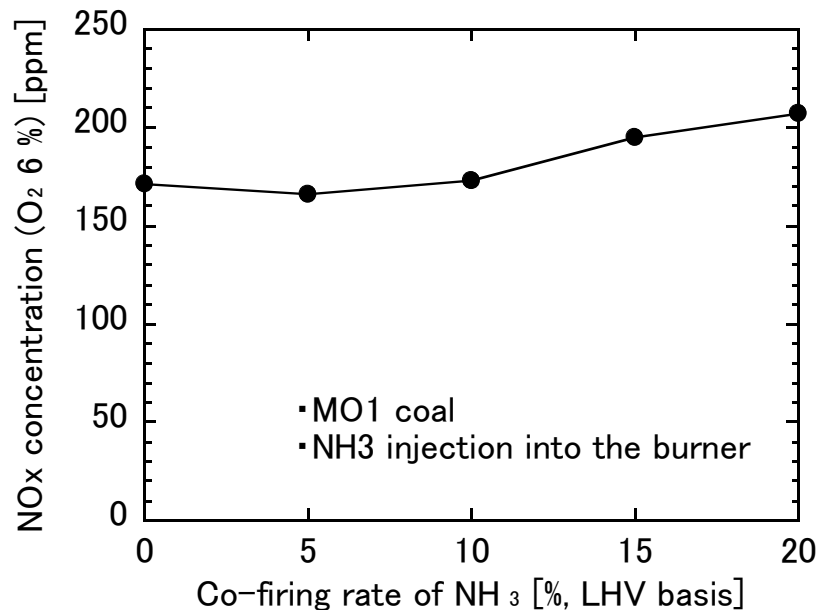
- ① Percentage of NH₃ blending (Max. 20%(LHV basis))
- ② Position of the side port where NH₃ is injected

Typical coal properties

Common bituminous coals were used for the experiment.

Item	Unit	Coal	
		MO1	MO2
Moisture (AD)	%	2.8	3.5
Ash (dry)	%	12.6	14.0
Volatile matter (dry)	%	35.4	36.0
Fixed carbon (dry)	%	52.1	50.1
Nitrogen (dry)	%	1.74	1.64
LHV (dry)	MJ/kg	29.2	28.4

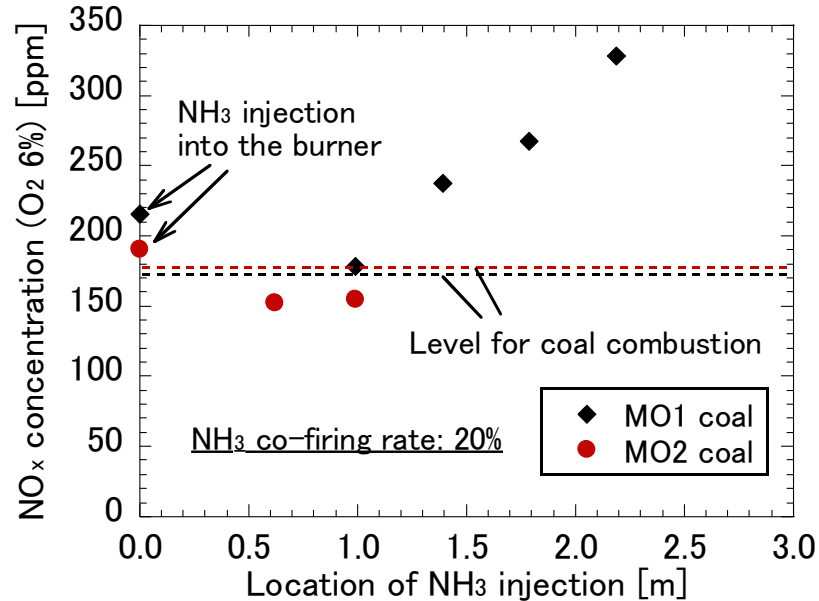
Effect of NH_3 co-firing rate to NO_x



When NH_3 was injected into the pulverized-coal burner

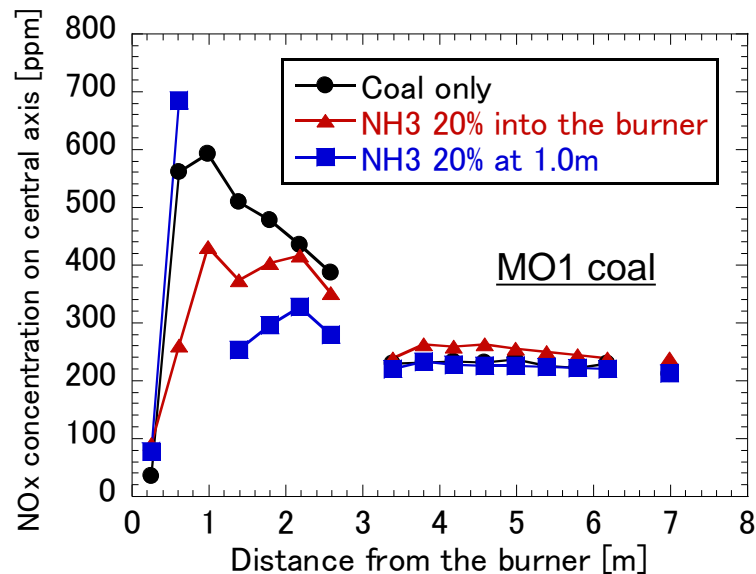
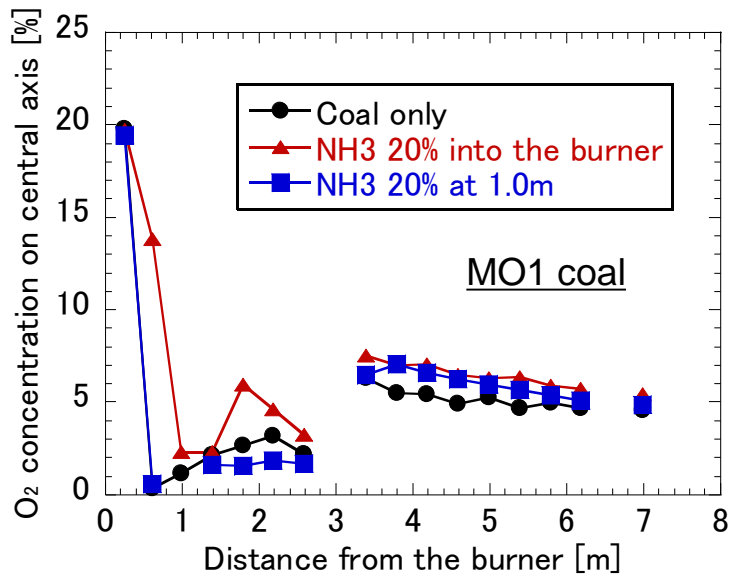
- Co-firing rate $\leq 10\%$ \Rightarrow Exhaust NO_x was much the same as coal combustion.
- Co-firing rate $\geq 10\%$ \Rightarrow Exhaust NO_x increased as input NH_3 increased.

Effect of NH₃ injection position



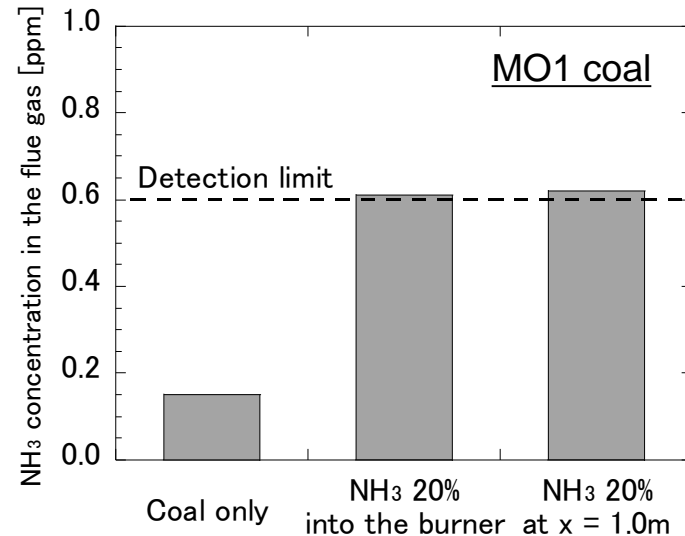
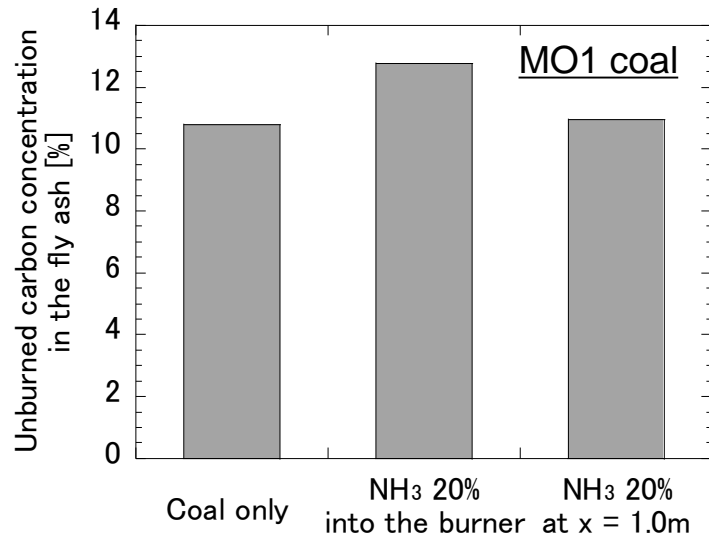
Exhaust NO_x concentration were decreased by injecting NH₃ through the side port, compared with injecting NH₃ into the coal burner.

Streamwise distributions of O₂ & NO_x



- In the case of NH₃ injection into the burner, **NO_x generation in the flame is low but it increases after the staged-air injection.**
- In the case of NH₃ injection at 1.0 m from the burner, **NO_x is decreased after NH₃ injection and is not much regenerated after the staged-air injection.**

Unburned contents

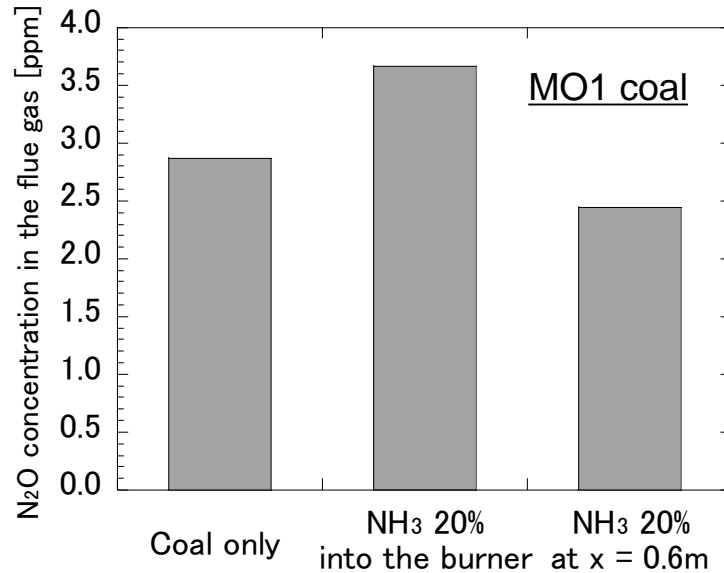


Unburned carbon concentration in the fly ash is slightly high in the case of NH₃ injection into the burner.

Unburned NH₃ concentration is increased by injecting NH₃, but is enough low considering the detection limit.

NH₃ injection through the side port has an advantage regarding the unburned carbon in the fly ash as well as NO_x.

Exhaust N_2O concentration



N_2O concentration was increased when NH_3 was injected into the burner, but is negligibly low to work as global warming gas.

Conclusions

Basic co-firing characteristics of pulverized coal and ammonia was investigated using single-burner test furnace. Main conclusions are below.

(1) Effect of NH_3 co-firing rate to NO_x generation in case of injecting NH_3 into the coal burner

- Co-firing rate $\leq 10\%$ \Rightarrow Exhaust NO_x was much the same as coal combustion.
- Co-firing rate $\geq 10\%$ \Rightarrow Exhaust NO_x increased as input NH_3 increased.

(2) Effect of position of the side NH_3 port

- When NH_3 was injected at 0.6 m or 1.0 m from the pulverized-coal burner, NO_x in the flue gas was decreased compared with the case injecting NH_3 into the coal burner.
- By injecting NH_3 into low O_2 , high NO_x region, fuel- NO_x generation is suppressed and NH_3 possibly works also as reductant for existing NO_x .

(3) Effect of NH_3 injection to unburned content

- When NH_3 was injected into the burner, unburned carbon in the fly ash slightly increased probably due to the lower flame temperature.
- Unburned NH_3 concentration was enough low even in the cases of NH_3 co-firing.

Acknowledgement

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Thank you for your attention!