



Ammonia Energy Association Presentation

AMMONIAC: A Chemical Looping-Based Process for Production of Green Ammonia

Laureate Professor Behdad Moghtaderi





Prof Moghtaderi's Group (Centre for Energy)

- Staff & Students
 - 20 Postdoctoral Research Fellows
 - 4 Technical & General support staff
 - 14 PhD Students
- Extensive array of laboratories and facilities worth \$22 M
- Grants Income: ≈ \$8 M/y
- Publications since 2017: 185
- RHD completions since 2013: 18



























Background

Liquid Hydrogen VS Liquid Ammonia

Characteristics and/or Properties	Liquid Hydrogen	Liquid Ammonia	Comment
Energy Content	9.9 MJ/Liter	15.3 MJ/Liter	An ammonia based system would be more compact and lighter
Boiling Point (@ 1 bar)	20K (-253 C)	240K (-33 C)	Given that any storage system will inevitably have some air with a boiling point of 79K, in the case of liquid hydrogen this air freezes to a solid with hazardous consequences.
Laten Heat of Vaporisation (@ 1 bar)	6.3 (kJ/m³) or 447 (kJ/kg)	2 (kJ/m³) or 1371 (kJ/kg)	The latent heat of liquid hydrogen in kJ/m³ is about three times bigger than that of Ammonia. Thus, for the same heat influx, the evaporation rate of liquid ammonia would be three times faster in terms of the volume of evaporated liquid.
Viscosity	13.06 (kg/m.s x 10 ⁶)	157.9 (kg/m.s x 10 ⁶)	The low viscosity of liquid hydrogen may lead to sloshing during transportation giving rise to vapour explosion if the sloshing liquid get in contact with the relatively warmer surfaces (e.g., roof of the storage tank).
Molecular Structure	Has Ortho and Para structures		Hydrogen is essentially a mixture of ortho- and parahydrogen. When liquefied at 20K, there is a slow but continuous transformation of the ortho-hydrogen to the lower energy parahydrogen in the form of boil-off which is undesirable. In addition, the co-existence of ortho- and para- rich liquid with different densities may lead to stratification and subsequent rollover with uncontrolled boiloff.

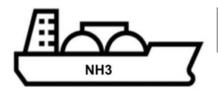




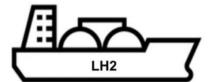
Why Ammonia might be one of the best bets for shipping low carbon fuel

~18% of H2 by mass

100% of H2 by mass



160,000 cbm LNG carrier \$3/kg H2 production cost



6-10 kWh/kg-NH3

\$0.48/kg-NH3

109,248 t

\$ 52.44 M

596.8 GWh

\$88/MWh

NH3 ← H2 → LH2

NH3 ← H2 → LH2

Total Cargo

685 kg/cbm

71 kg/cbm

Total Cost Total Energy

Specific cost of energy

12 kWh/kg-LH2

\$7.15/kg -LH2

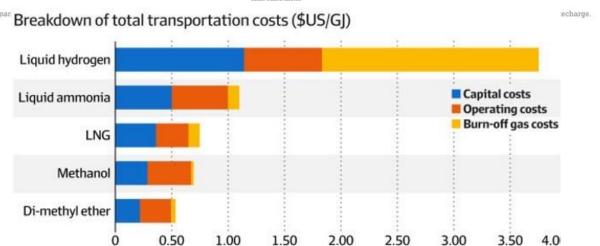
11,376 t

\$81.34 M

404.8 GWh

\$200/MWh

NH3: Ammonia LH2: Liquid Hydroge cbm: cubic meter





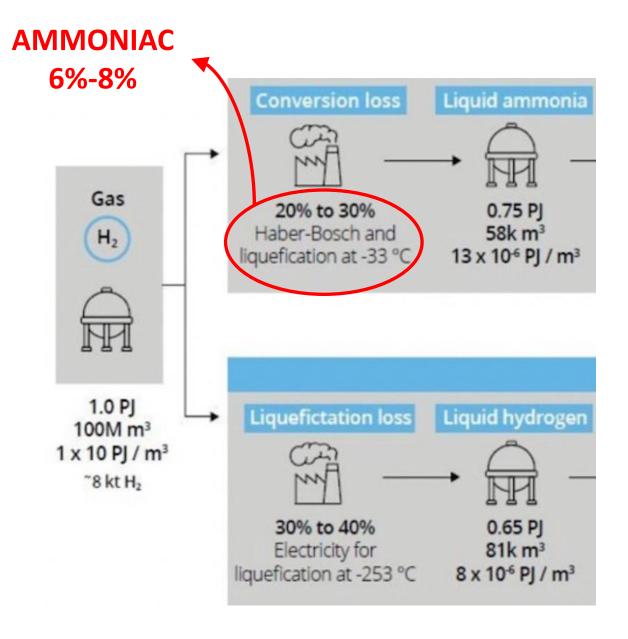


The Suiso Frontier left Hastings (Vic) in Jan 2022 after loading its cargo of liquid hydrogen.









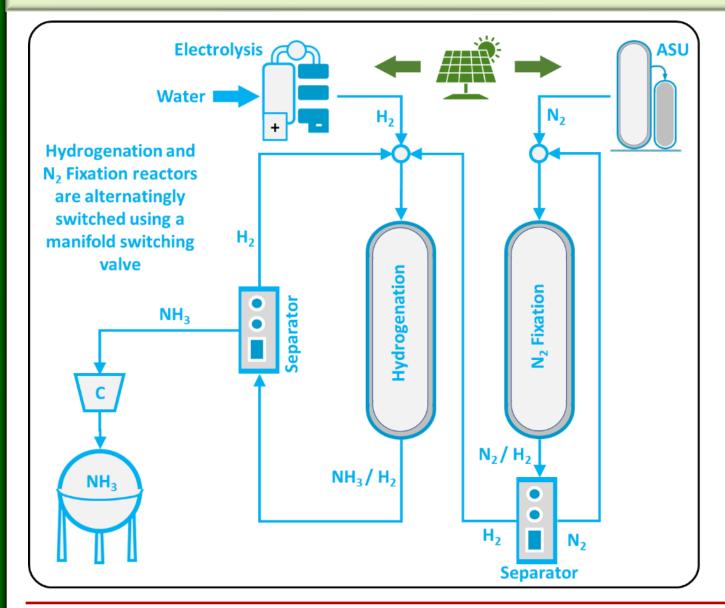
- Since 1913 the production of ammonia at industrial scale has been based on the Haber-Bosch process which is considered as one of the most important inventions of the 20th century.
- The Haber-Bosch process, N2+3H2→2NH3, requires a catalyst to promote the reaction between nitrogen and hydrogen, and is only feasible at high temperatures (~430°C) and pressures (~100 atm).
- The process has been extremely well optimised over the past 110 years.
- However, given that the primary source of hydrogen in the Haber-Bosch process is natural gas, the process is still responsible for approximately 1.1% of the global energy consumption and about 1% of global greenhouse gas (GHG) emissions.
- Typically, 1.9 tonnes of CO2 are released per tonne of ammonia produced when natural gas is utilised to produce hydrogen.







AMMONIAC Process



- The AMMONIAC process is a thermochemical looping process for production of green ammonia from renewable hydrogen and atmospheric nitrogen.
- The process essentially "breathes in" nitrogen and "breathes out" ammonia by simply cycling between different reaction environments at ambient pressure.
- This breakthrough process is a simple, cost effective and environmentally friendly alternative to the conventional Haber-Bosch process and other emerging technologies for ammonia synthesis.



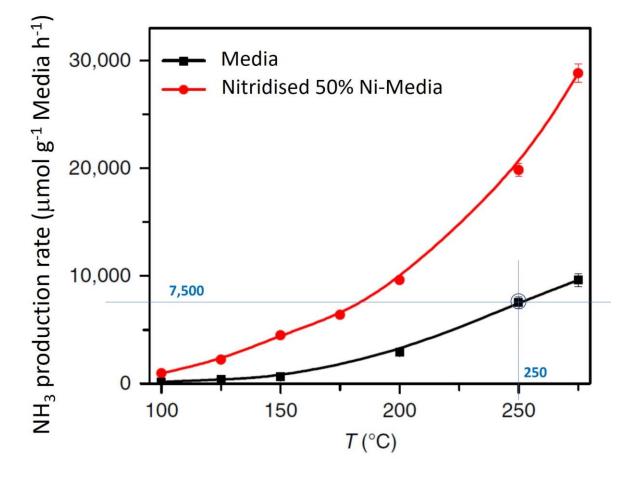








TRL 4

















AMMONIAC Project



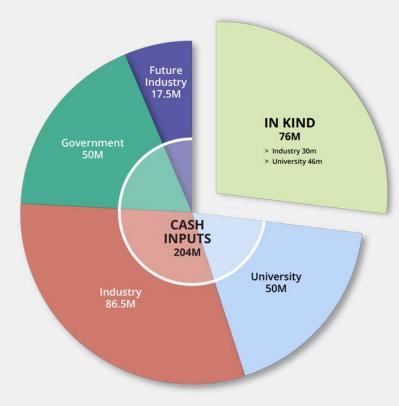






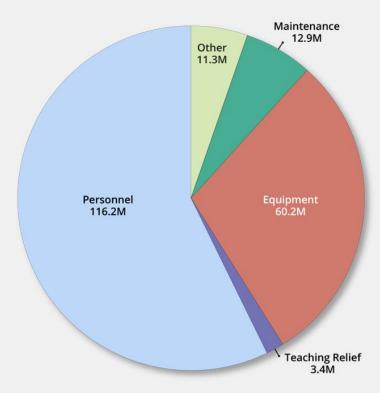


Leveraged government funding



\$280M program

(government funding leveraged at 4.5 to 1)

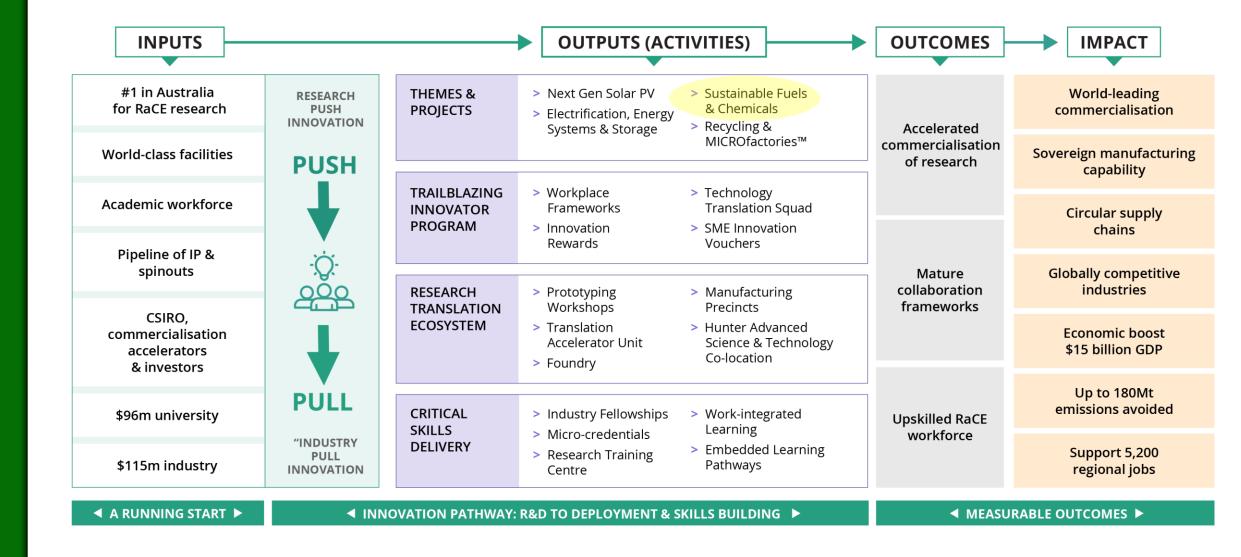


Cash expenditure \$204M















Theme 3: Sustainable Fuels and Chemicals Manufacturing

Theme Leader: Prof Behdad Moghtaderi

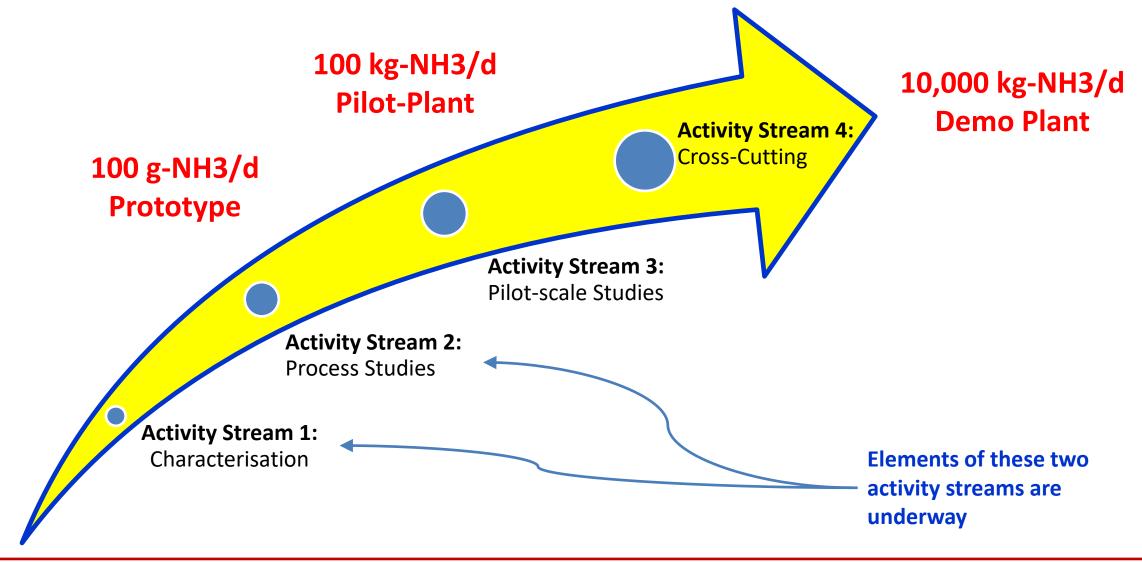
- AMMONIAC: A Chemical Looping-Based Process for Production of Green Ammonia (Element 1 Pty Ltd)
- Achieving Negative Emissions in Production of Green Steel and Green Chemicals Using the VAMCO Family of Gas Separation Technologies (Ascon Energy Pty Ltd)
- KIMIYA: A Technology Platform for Conversion of Organic Waste to Sustainable Chemicals and Fuels (ELMNTRE Pty Ltd)
- Development of a Novel, Low-Cost, High Performance, Safe and Sustainable Hydrogen Storage Material (LAVO Hydrogen Storage Technology Pty Ltd)
- Advanced Technology Hydrogen Compressor Development and Testing (Siemens Energy Pty Ltd)







Project Has Commenced as of April 5, 2023







Questions

CRICOS Provider 00109J | www.newcastle.edu.au





