

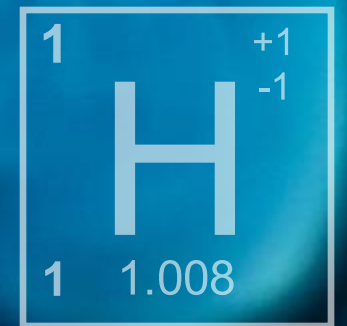


# Carbon Footprint Methodology

AEA Annual Conference

November 13-15

BRUSELAS • PARÍS • ROTTERDAM • WASHINGTON D.C.  
• SANTIAGO • BOGOTÁ



# Agenda

- ▶ CCUS & Emissions Reductions
  - ▶ Definitions
  - ▶ Straightforward Approach
  - ▶ Dynamic Plant Splitting Approach
  - ▶ Variable CCS & Urea
- ▶ Handling Multifunctionality & Allocation Principles
  - ▶ Steam
  - ▶ Solid Carbon (Methane Pyrolysis)

1

# CCUS Definitions & Emissions Reduction

# Carbon Capture & Storage Definition

The AEA scheme allows for emissions reductions with long-term CCS

---

## ▶ Carbon Capture & Storage

- ▶ Captured CO<sub>2</sub> is CO<sub>2</sub> that is extracted from the process and then available for conditioning and storage
- ▶ After capture, CO<sub>2</sub> can be transported via pipeline to underground geological storage sites
- ▶ It is assumed that **the assured geological storage period to be considered long-term will be 100 years<sup>1</sup>**, although this will be subject to local regulatory requirements applicable to the participant
- ▶ **Permanently storing CO<sub>2</sub> resulting from ammonia production reduces the CFP of ammonia**

<sup>1</sup> IPCC, 2005: IPCC Special Report on Carbon Dioxide Capture and Storage. Prepared by Working Group III of the Intergovernmental Panel on Climate Change [Metz, B., O. Davidson, H. C. de Coninck, M. Loos, and L. A. Meyer (eds.)]. Cambridge University Press, Cambridge, United Kingdom and New York, NY, USA, 442 pp. P. 197

# Carbon Capture & Utilization Definition

The AEA scheme allows for emissions reductions with only long-term CCU

## ▶ Carbon Capture & Utilization

- ▶ Once carbon is captured, it can be subsequently utilized in various commercial products

### Long-term (30+ years).

#### Long-Term CCU Explained

- ▶ Long-term CCU: Permanent or semi-permanent sequestration of CO<sub>2</sub>.
- ▶ Example: Transformation of CO<sub>2</sub> into a mineral carbonate
- ▶ **Eligible for emissions reduction due to long-term CO<sub>2</sub> storage.**

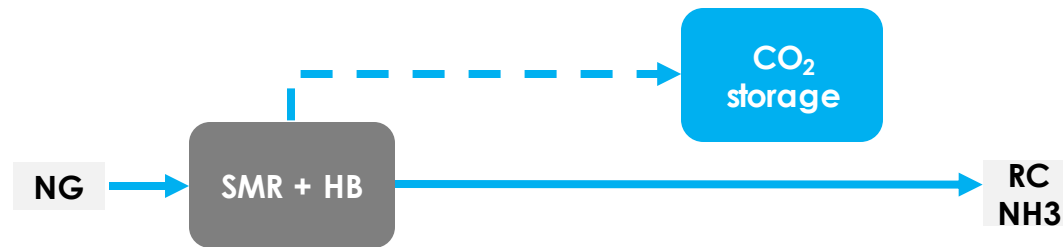
### Short-term (<30 years)

#### Short-Term CCU Explained

- ▶ Short-term CCU: Temporary storage of CO<sub>2</sub> in products with a lifespan of less than 30 years.
- ▶ Example: Urea for fertilizer – CO<sub>2</sub> released when applied to fields.
- ▶ **No emissions reduction due to temporary carbon sequestration.**

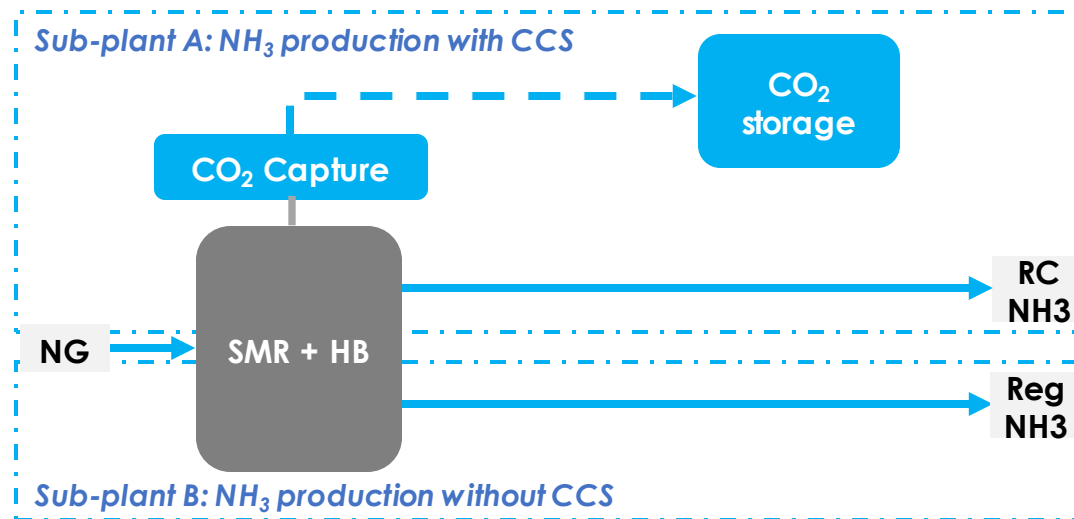
# Straightforward & Dynamic Plant Splitting Approach

## Two approaches to calculate the emissions reduction from CCS



### Straightforward LCA Approach

CCS reduces the footprint of all the ammonia produced



### Dynamic Plant Splitting Approach

CCS reduces only part of the ammonia produced, and the rest keeps the regular footprint

#### Conditions for Applying DPS:

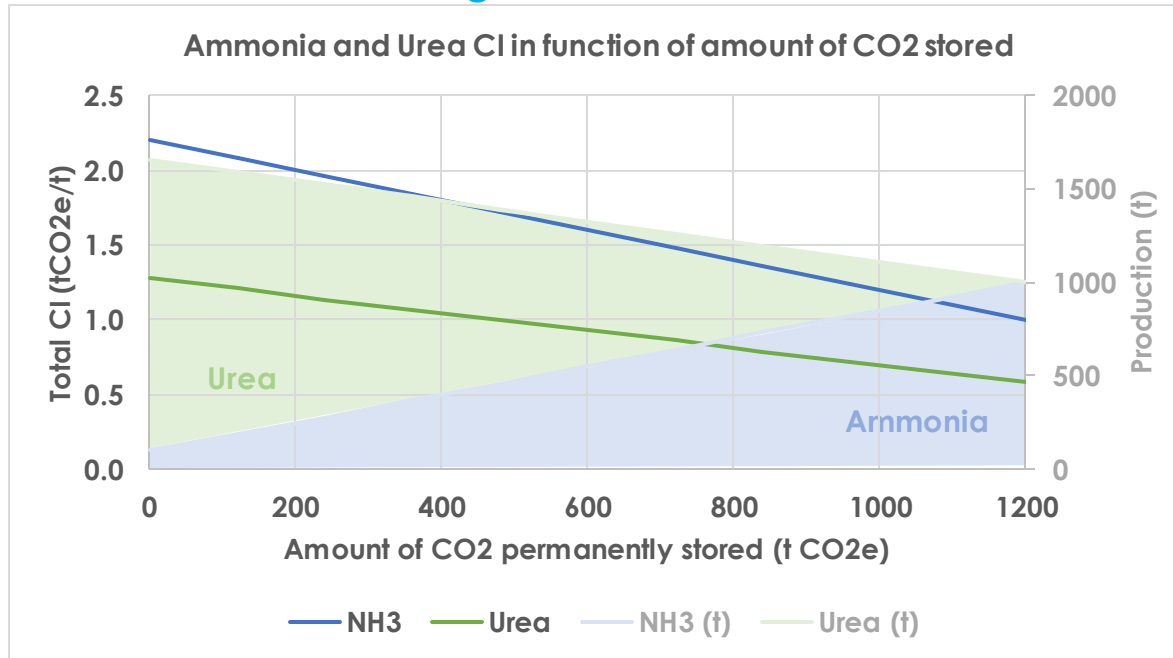
- Ammonia production plants that apply CCS which has fluctuating CO<sub>2</sub> sequestration rates over time.

# Accounting for CCS in case of multiple products

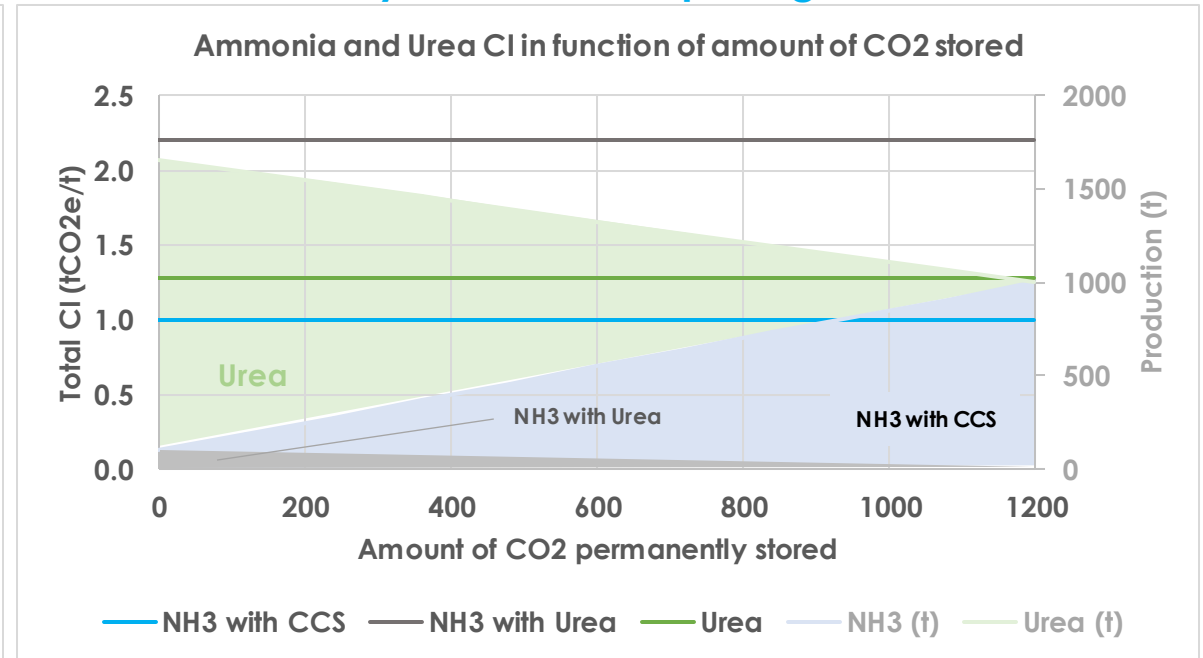
When coproducing urea, there are at least two ways to assign the emissions reduction

The examples below assume that all the CO<sub>2</sub> not used in urea, is geologically stored.

## Straightforward LCA



## Dynamic Plant Splitting



The total CFP of both products decrease in proportion to the amount of avoided CO<sub>2</sub> emissions due to CCS.

However, with dynamic plant splitting, an amount of low-carbon NH<sub>3</sub> is produced in proportion to the amount of avoided CO<sub>2</sub> emissions due to CCS.

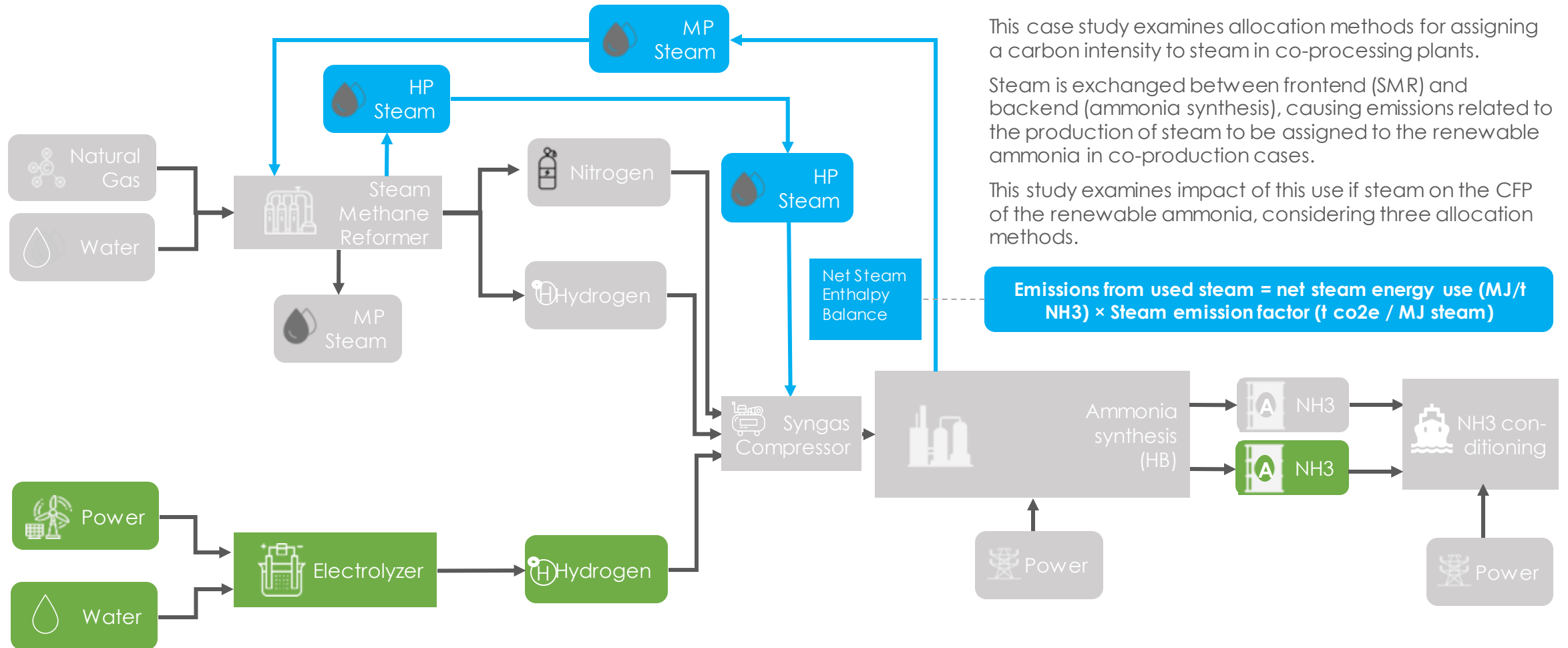
# 2

## Handling Multifunctionality & Allocation Principles



# Emissions From Steam Used to Produce Renewable NH<sub>3</sub>

Steam exported from conventional SMR used to compress H<sub>2</sub> and N<sub>2</sub> for renewable NH<sub>3</sub>



This case study examines allocation methods for assigning a carbon intensity to steam in co-processing plants.

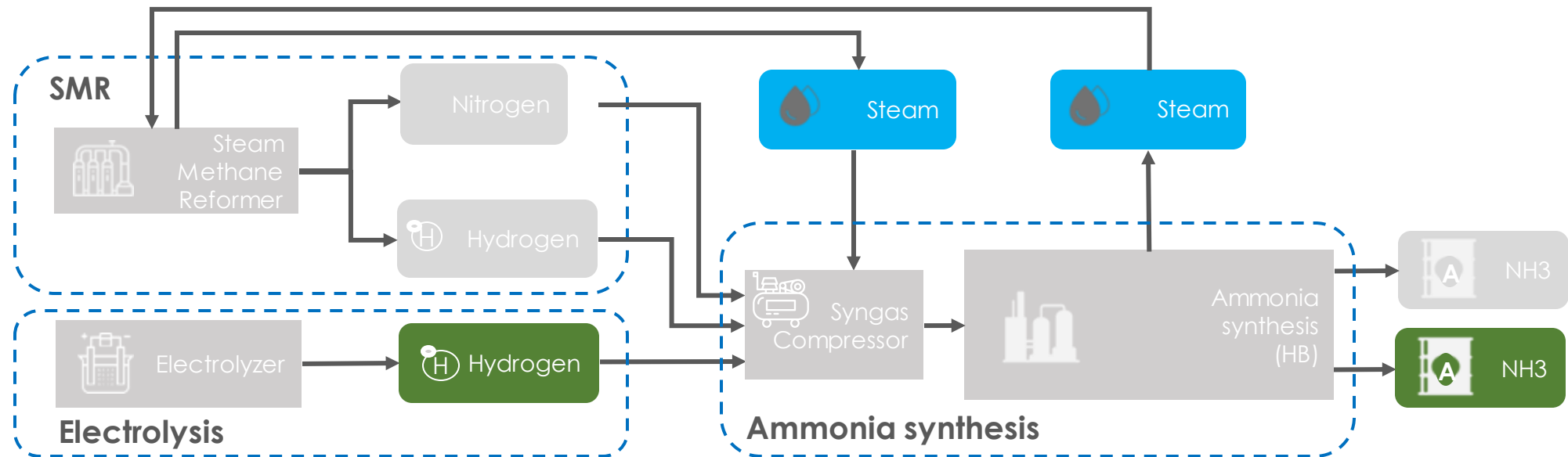
Steam is exchanged between frontend (SMR) and backend (ammonia synthesis), causing emissions related to the production of steam to be assigned to the renewable ammonia in co-production cases.

This study examines impact of this use if steam on the CFP of the renewable ammonia, considering three allocation methods.

$$\text{Emissions from used steam} = \text{net steam energy use (MJ/t NH}_3\text{)} \times \text{Steam emission factor (t co}_2\text{e / MJ steam)}$$

# Assigning a CFP to Co-Produced Steam

Different variants of energy-based allocation methods can be applied



## Emissions Allocated to Net Steam Consumption

1) Energy losses fully assigned to main reformer products (hydrogen)

Steam is only assigned the emissions of heat input required to generate the steam in a heat exchanger, considering 98% efficiency

2) Steam also carries emissions related to process energy losses

Steam is allocated the upstream and process emissions in proportion to its energy content (steam energy content vs hydrogen energy content)

3) System expansion considering that the steam replaces steam from a boiler

Steam is allocated the emissions of heat input required to generate the steam in a natural gas fired boiler, considering 90% efficiency

# Steam CFP Varies Slightly By Approach Used

## ... and the ammonia CFP also varies slightly

Example shows small variations in steam CFP when applying the three different options of allocation methods.

	Option 1	Option 2	Option 3
Description of Approach	Steam is only assigned the emissions of heat input required to generate the steam in a heat exchanger, considering 98% efficiency	Steam is allocated the upstream and process emissions in proportion to its energy content	Steam is allocated the emissions of heat input required to generate the steam in a natural gas fired boiler, considering 90% efficiency
<b>Steam CFP</b> in g CO <sub>2</sub> e/MJ	<b>67.2</b> (fixed value)	<b>73.5</b> (process dependent)	<b>73.2</b> (fixed value)
<b>Renewable Ammonia</b> in kg CO <sub>2</sub> e/kg	<b>0.356</b>	<b>0.372</b>	<b>0.371</b>

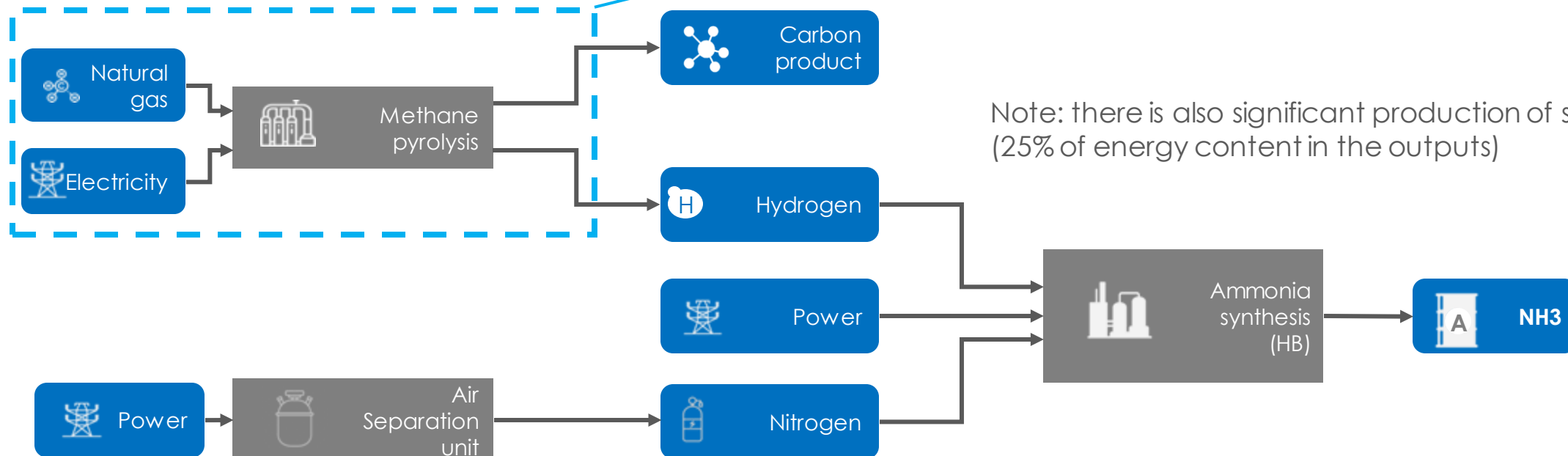
# Allocating Emissions using Methane Pyrolysis for H<sub>2</sub> & NH<sub>3</sub>

## Ammonia CFP depends emissions allocation to the Carbon and H<sub>2</sub>

Different carbon products possible:

- ▶ Carbon black
- ▶ High purity carbon
- ▶ Carbon

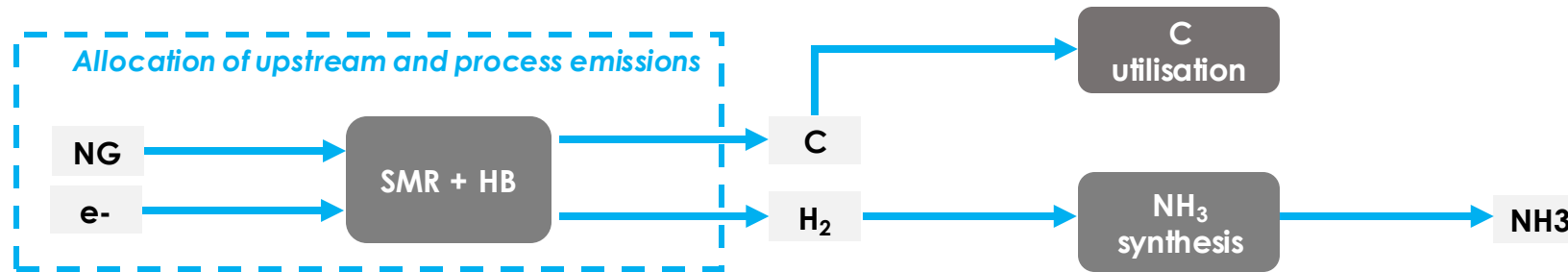
**How to allocate the upstream and process emission of methane pyrolysis to the two co-products: the carbon product (e.g., carbon black) and hydrogen?**



Note: there is also significant production of steam (25% of energy content in the outputs)

# Allocation Methods Assessed

...following allocation method hierarchy specified by ISO 14067



Application subject to various conditions

## System expansion by substitution

Emissions assigned to hydrogen based on emissions of reference hydrogen production

Which reference production would H<sub>2</sub> from NG pyrolysis substitute?

→ **Not considered applicable here**

## Physical causality based allocation

Impact on pyrolysis process emissions of making H<sub>2</sub> available for use assigned to the H<sub>2</sub>

Emissions assigned based on the change of emissions caused by making an extra unit of co-product available (keeping other co-product quantities constant)

## Energy based allocation A

Emissions assigned to co-products based on their respective energy content

Energy content of the products is made up by the energy content of the inputs  
Always assigns the same process emissions per MJ to all the co-products

Variant

## Energy based allocation B

Emissions assigned to co-products based on their respective energy content, with adjustment for carbon content

Considers difference in emissions from separate production

## Mass based allocation

Emissions assigned to co-products based on their respective mass

Mass of the products is made up by the mass of the inputs  
Not suitable for steam  
Generally, not considered suitable for H<sub>2</sub> due to its low molecular weight

## Value based allocation

Emissions assigned to co-products based on their respective economic value

Value is a fundamental basis for splitting the emissions between the products

## No allocation – carbon is stored

The process yields only one product: hydrogen, therefore allocation is not needed

When hydrogen production by pyrolysis will yield more carbon than needed by the applications, the carbon will need to be permanently stored

# Impact of Allocation Methods on CFP of Hydrogen

## Significantly varying results for different methods of allocation

